An Active, General, and Long-Lived Palladium Catalyst for Cross-Couplings of Deactivated (Hetero)aryl Chlorides and Bromides with Arylboronic Acids

Takashi Hoshi,* Tomonobu Honma, Ayako Mori, Maki Konishi, Tsutomu Sato, Hisahiro Hagiwara, and Toshio Su[zu](#page-10-0)ki

Department of Chemistry and Chemical Engineering, Niigata University, 8050, 2-Nocho, Ikarashi, Nishi-ku, Niigata 950-2181, Japan

ABSTRACT: An active, general, and long-lived palladium catalyst for Suzuki−Miyaura reactions of aryl and heteroaryl chlorides deactivated by steric hindrance, electron richness, and coordinating functional groups is reported. In reactions of arylbromide bearing two *o-tert-*butyl substituents, C(sp³)−H arylation of the tert-butyl group, rather than the Suzuki−Miyaura reaction, proceeded in excellent yield. The key to the success of the reactions was the development of biphenylene-substituted dicyclohexylruthenocenylphosphine (CyR-Phos) as a supporting ligand.

■ INTRODUCTION

Since the discovery of the beneficial effects of the using $P(t-$ Bu)₃ as a supporting ligand for palladium-catalyzed Buchwald− Hartwig amination of cheap but unreactive aryl chlorides, $¹$ the</sup> development of a number of ligands that greatly improve catalytic performance has expanded the substrate sco[pe](#page-11-0) of palladium-catalyzed cross-coupling reactions.^{2−4} Recently, we reported biphenylene-substituted di-tert-butylruthenocenylphosphine, R-Phos (Figure 1, left), as an e[ff](#page-11-0)e[ct](#page-11-0)ive ligand for

Figure 1. Biphenylene-substituted ruthenocenylphosphine ligands: R-Phos and CyR-Phos.

sterically difficult Suzuki-Miyaura reactions.⁵ That work was highlighted by the facile construction of highly crowded tetraortho-substituted biaryls from aryl chlorides. [H](#page-11-0)ere we report an active, general, and long-lived palladium catalyst for Suzuki− Miyaura reactions of aryl and heteroaryl chlorides using the dicyclohexylphosphino analogue, CyR-Phos (Figure 1, right). The catalyst generated from CyR-Phos enabled a broad substrate scope, low catalyst loading, and short reaction times in constructing tri-ortho-substituted biphenyls from aryl chlorides bearing two ortho-substituents (methyl, ethyl, and isopropyl) and a structurally diverse set of ortho-substituted arylboronic acids. The catalyst was also highly active in the

reaction of further sterically hindered arylbromide bearing two o-tert-butyl substituents: C(sp³)−H arylation of the tert-butyl group, rather than the Suzuki−Miyaura reaction, proceeded in excellent yield. The high catalytic performance was maintained in the reactions of aryl chlorides deactivated by electrondonating methoxy and amino groups. Tolerance toward the amino group was also observed in pyridyl and pyrimidyl chlorides bearing an unprotected amino group, which typically must be protected to prevent catalytic activity loss due to strong coordination to the catalyst.⁶

RESULTS A[N](#page-11-0)D DISCUSSION

Initially, we examined the precatalyst generated from CyR-Phos in the reaction of 2,6-dimethylchlorobenzene (1a) with 2 methylphenylboronic acid (2a) (Table 1). The ligand/Pd ratio was examined in a range from 1:1 to 5:1, and the best result was obtained at a ratio of 4:1 (Table 1, [en](#page-1-0)tries 1−4). However, when using precatalyst solution that had been stored in a freezer (ca. -20 °C) for 5 days, a s[ig](#page-1-0)nificant drop in yield was observed (Table 1, entry 5). When the precatalyst solution was prepared in dioxane instead of THF, the decrease in yield after storage was imp[ro](#page-1-0)ved but could not be completely prevented (Table 1, entries 6 and 7). In our previous work using R-Phos, the precatalyst could be used even after storage for several month[s w](#page-1-0)ithout any loss of catalytic performance. To maintain the quality of the CyR-Phos precatalyst during storage, we further optimized the precatalyst and found that using an aryl

Received: September 20, 2013 Published: October 27, 2013

Table 1. Optimization of Precatalyst Based on CyR-Phos^a

^a All reactions were run with 1.0 mmol of 1a and 1.5 equiv of 2a. $b_{\text{Prec} \rightarrow \text{t}}$ Precatalyst solution was prepared in THF. ^cPrecatalyst solution was prepared in dioxane. d_2 h.

bromide additive permits the storage of the precatalyst. In addition, the yield of the catalytic reaction was substantially improved as compared to the reaction when the precatalyst was prepared without aryl bromide additive. Although we cannot presently explain the beneficial effects of aryl bromide on both the storage and activation of the precatalyst, λ the reaction using the precatalyst prepared in the presence of ethyl 4 bromobenzoate, for example, went to co[m](#page-11-0)pletion (Table 1, entry 8) and the catalytic performance remained unchanged after 20 days (Table 1, entry 9). Using 0.01 mol % Pd, the reaction proceeded in 87% yield in 2 h, corresponding to a turnover number of 8,700 and a turnover frequency of 4,350 h⁻¹ (Table 1, entry 10). To the best of our knowledge, these are the highest turnover number and turnover frequency reported for this sterically demanding Suzuki−Miyaura reaction.3f,o,v

With an efficient precatalyst in hand, we explored the substrat[e sco](#page-11-0)pe. Using 0.025 mol % of Pd, the reactions of 1a with more sterically hindered o-alkyl- or o-aryl-substituted arylboronic acids 2b−d gave the corresponding products 3ab− ad in good to excellent yields (Table 2, entries 2−4). Also, a sterically diverse set of o-alkoxy-substituted arylboronic acids 2e−g coupled with 1a to provide the biphenyls 3ae−ag in comparable yields under identical conditions (Table 2, entries 5−7). Though a higher catalyst loading (0.1 mol %) was required, R-Phos was also effective for the same substrate combinations (Table 2, entries 8−14).

The high catalytic performance regardless of the size and coordinating functional group of arylboronic acids motivated us to examine CyR-Phos and R-Phos in reactions of more sterically hindered aryl chlorides with the same set of arylboronic acids. In reactions of 2,6-diethylchlorobenzene (1b), the efficacy of CyR-Phos was maintained to afford the coupling products 3ba−bg in excellent yields in <2 h with only 0.05 mol % Pd (Table 3, entries 1−7). In contrast, the catalyst generated from R-Phos required 20 times higher loading (1 mol %) to provide comparable outcomes from unfunctionalized arylboronic acids 2a−d (Table 3, entries 8−11). When omethoxy-substituted arylboronic acid 2e was used as a coupling Table 2. Suzuki−Miyaura Reactions of 2,6- Dimethylchlorobenzene (1a) with ortho-Substituted Arylboronic Acids^a

^a All reactions were run with 1.0 mmol of aryl chloride and 1.5 equiv of arylboronic acid. $b_{0.025}$ mol % $Pd(dba)_2$, 0.1 mol % CyR-Phos, 0.5 mol % 4 -BrC₆H₄CO₂Et. ^c0.1 mol % Pd(dba)₂, 0.3 mol % R-Phos.

Table 3. Suzuki−Miyaura Reactions of 2,6- Diethylchlorobenzene (1b) with ortho-Substituted Arylboronic Acids^a

Et		Et
1b	$Pd(dba)_{2}/L$	Et R $3ba$: R = Me
+ $(HO)_2B$ 2a-2g	3 equiv K ₃ PO ₄ ·H ₂ O dioxane, 100 °C	$3bb: R = Et$ 3bc : $R = i$ -Pr 3bd : $R = Ph$ $3be$: R = MeO $3bf$: R = EtO 3 bg: R = i-PrO

^a All reactions were run with 1.0 mmol of aryl chloride and 1.5 equiv of arylboronic acid. b 0.05 mol % $Pd(dba)$, 0.2 mol % CyR-Phos, 1.0 mol $\%$ 4-BrC₆H₄CO₂Et. ^c1 mol % Pd(dba)₂, 3 mol % R-Phos.

partner, the coupling product 3be was obtained in only modest yield (Table 3, entry 12). Though we have no explanation for the remarkably detrimental effect of the coordinating methoxy group on the activity of R-Phos, the clear superiority of CyR-Phos over R-Phos for the highly hindered chloride 1b is likely due to decreased steric demand of the dialkylphosphino group, which would increase the reactivity of the congested catalytic species.

CyR-Phos was also effective for coupling further hindered 2,6-diisopropylchlorobenzene (1c) with arylboronic acids 2a−g using only the same low catalyst loading (Table 4, entries 1−7).

Table 4. Suzuki−Miyaura Reactions of 2,6- Diisopropylchlorobenzene (1c) with ortho-Substituted Arylboronic Acids^a

^a All reactions were run with 1.0 mmol of aryl chloride and 1.5 equiv of arylboronic acid. ^bYields of homocoupling byproduct (left) and 1,3diisopropylbenzene (right).

To the best of our knowledge, 1c is the most hindered aryl chloride that has been successfully employed in the Suzuki− Miyaura reaction; to date, only two catalysts have been reported to be effective for this chloride.^{4f,8} Also worth noting is that significant amounts of the homocoupling byproducts from arylboronic acids and 1,3-diisopropylb[enze](#page-11-0)ne, the dehalogenation product of 1c, formed as the size of the boronic acids increased (Table 4, entries 1−4 and 7). We speculate that after the oxidative addition step, the resulting arylpalladium intermediate bearing two o-isopropyl substituents underwent the aryl−aryl exchange with arylboronic acid to give the monoortho-substituted arylpalladium complex and 2,6-diisopropylphenylboronic acid (Scheme 1); then, the former underwent the common transmetalation with another arylboronic acid to give the homocoupling byproduct with regeneration of the catalyst, and the latter underwent the protodeboronation to give the net dehalogenation product. The remarkable replacement of the common aryl−chloride exchange with the unusual aryl−aryl exchange in the reaction of relatively hindered boronic acids is likely due to the avoidance of forming the highly congested diarylpalladium intermediate.⁹ The formation of the boronic acid-derived arylpalladium intermediate is also sterically favorable due to the significa[nt](#page-11-0) Scheme 1. Aryl−Aryl Exchange between Arylpalladium Intermediate and Arylboronic Acid

reduction of the steric congestion of the arylpalladium intermediate derived from the bulky aryl chloride 1c.

In contrast to the excellent activity in the reactions of 2,6 dialkylchlorobenzenes 1a−c, the catalyst could not mediate the coupling of the most sterically hindered 2,4,6-tri-tert-butylchlorobenzene (1d) even with 4-methoxyphenylboronic acid (2h), the less hindered coupling partner with no orthosubstituents (Table 5, entry 1). Fortunately, however, the

Table 5. C (sp^3) –H Arylation on *o-tert-*Butyl Substituent a

^a All reactions were run with 1.0 mmol of aryl chloride and 1.5 equiv of arylboronic acid. $b_{0.025}$ mol % $Pd(dba)₂$, 0.1 mol % CyR-Phos, 0.5 mol % $4-BrC_6H_4CO_2Et$. \cdot 0.05 mol % $Pd(dba)_2$, 0.2 mol % CyR-Phos, 1.0 mol % $4\text{-}BrC_6H_4CO_2Et$.

reaction of the bromide counterpart 1e with 2h proceeded to completion with only 0.025 mol % Pd (Table 5, entry 2). Though bromides are well-known as much more reactive halides compared with chlorides in palladium-catalyzed crosscoupling processes, the facile progress of the reaction highlights the excellent tolerance of the catalyst for sterically hindered aryl halides in the oxidative addition step. In addition, neither the Suzuki−Miyaura reaction nor the homocoupling side reaction, but rather the $C(sp^3)$ -H arylation of the o-tert-butyl group took place exclusively.¹⁰ The lack of any C(sp²)–C(sp²) bondforming processes suggests that transmetalation between the bromide-derived aryl[pal](#page-11-0)ladium intermediate and 2h is completely inhibited by the steric hindrance of two large o-tert-butyl substituents. On the other hand, the tremendous congestion in the palladium complex allows either of the two tert-butyl substituents to stay continuously in close contact with Pd,

which can facilitate the intramolecular activation of the inert $C(sp^3)$ –H bond (Scheme 2). The resulting much less hindered

Scheme 2. Intramolecular C(sp^3)–H Activation in Bromide-Derived Arylpalladium Intermediate

alkylpalladium intermediate undergoes transmetalation with 2h followed by reductive elimination to give the $C(sp^3)$ -H arylation product 4eh. Using 0.05 mol % Pd, the couplings of 1e with various ortho-substituted arylboronic acids 2a−g also gave the $C(sp^3)$ -H arylation products 4ea-4eg in excellent yields (Table 5, entries 3−8).

The ability of the catalyst to couple with aryl chlorides deactivated [by](#page-2-0) electron-donating groups is also of great importance in palladium-catalyzed coupling reactions. Thus, we proceeded to assess the efficiency of CyR-Phos in the reactions of aryl chlorides containing electron-donating $group(s)$ (Table 6). Using 0.025 mol % Pd, the reaction of

Table 6. Suzuki−Miyaura Reactions of Electronically Deactivated Aryl Chlorides^a

^a All reactions were run with 1.0 mmol of aryl chloride, 1.5 equiv of arylboronic acid, and 3 equiv of $K_3PO_4·H_2O$ in dioxane at 100 °C. $b_{0.025}^{b_{0.025}}$ mol % Pd(dba)₂, 0.1 mol % CyR-Phos, 0.5 mol % 4- $BrC_6H_4CO_2Et.$ co.25 mol % $Pd(dba)_2$, 1.0 mol % CyR-Phos. 5.0 mol % $4-BrC_6H_4CO_2Et.$ d) and % Pd(dba)₂, 0.4 mol % CyR-Phos, 2.0 mol % 4- BrC_6H_4t -Bu.

4-chloroanisole (1f) with 4-methylphenylboronic acid (2i) went to completion (Table 6, entry 1). The catalyst was also tolerant of a coordinating group at the ortho position (Table 6, entry 2). Although the reaction of 2,4-dimethoxychlorobenzene (1h) proceeded in only 19% yield due to strong deactivation by two methoxy groups, high yield was obtained at higher catalyst loading (Table 6, entries 3 and 4). Using 0.1 mol % Pd, the reaction of 4-chloroaniline (1i) with 2h proceeded in 70% yield (Table 6, entry 5).

The high yield from the reaction of 4-chloroaniline (1i) raised the prospect that CyR-Phos would be effective for coupling other aryl and heteroaryl chlorides bearing an amino group. Nitrogen heteroaromatics are ubiquitous substructures in biologically active compounds and advanced materials.¹¹ However, in the Suzuki−Miyaura coupling of pyridyl and pyrimidyl chlorides bearing an unprotected amino gro[up,](#page-11-0) substantial catalyst deactivation is often observed due to coordination with palladium.¹² To our delight, the use of CyR-Phos with 0.1 mol % Pd allowed for aminopyridyl or aminopyrimidyl chlorides t[o c](#page-11-0)ouple with arylboronic acids in high yield; the one exception was the most strongly basic coupling partner, 4-amino-2-chloropyridine (1m) (Table 7,

Table 7. Suzuki−Miyaura Reactions of Pyridyl and Pyrimidyl Chlorides Bearing Amino and Hydroxyl Groups^a

a
All reactions were run with 1.0 mmol of heteroaryl chloride, 1.5 equiv of arylboronic acid, 0.1 mol % $Pd(dba)_2$, 0.4 mol % CyR-Phos, 2.0 mol % PhBr, and 3 equiv of $K_3PO_4·H_2O$ at 100 °C. b Reaction was run in tert-amyl alcohol. ϵ Reaction was run in dioxane.

entries 1−6). As can be seen from the reactions using simple aryl chlorides, the catalytic activity was high for this substrate class, regardless of the steric character. In fact, 5-chloro-2,6 dimethyl-4-pyrimidinamine (1o), which possesses two bulky substituents ortho to the chlorine, coupled with 2a to afford the corresponding tri-ortho-substituted heterobiaryl 3oa in 72% yield (Table 7, entry 6). An unprotected hydroxypyridyl chloride $(1p)$ was also well-tolerated in the catalytic system (Table 7, entry 7).

■ CONCLUSION

In summary, we have demonstrated that the palladium complex generated from CyR-Phos is an active, general, and long-lived catalyst for Suzuki−Miyaura reactions of aryl and heteroaryl chlorides deactivated by steric hindrance, electron richness, and coordinating functional groups. The excellent tolerance of the catalyst toward steric hindrance in the oxidative addition step could lead to the highly congested arylpalladium intermediate whose steric hindrance caused the common aryl−chloride exchange to be replaced by the unusual but sterically more

favorable aryl−aryl exchange in the subsequent transmetalation with boronic acid. In the reaction of arylbromide bearing two o tert-butyl substituents, the close contact between the tert-butyl group and the palladium in the arylpalladium intermediate allowed intramolecular $C(sp^3) - H$ activation followed by arylation with boronic acid in excellent yield. The precatalyst could be stored without any loss of catalytic performance through the use of an aryl bromide additive, which also significantly improved the yield of the coupling reaction. The superiority of CyR-Phos over R-Phos in forming tri-orthosubstituted biphenyls demonstrated that the dialkylphosphino substructure is an important control element for the efficiency of the ligands. Our ongoing efforts are directed toward applying CyR-Phos to additional palladium-catalyzed cross-coupling processes. The development of more effective ligands is also currently underway in our laboratory.

EXPERIMENTAL SECTION

General. All reactions were carried out under an atmosphere of argon in flame-dried glassware with a magnetic stir bar. Flash chromatography was performed with silica gel 60 (230−400 mesh). ¹ 1 H NMR spectra were recorded at 270.1 MHz, 399.9 MHz, 499.9 MHz, or 699.8 MHz. 13C NMR were recorded at 100.6 MHz, 125.7 MHz, or 176.0 MHz. The ¹H chemical shifts were referenced to internal Me₄Si (δ 0.00 ppm). The ¹³C chemical shifts were referenced to CDCl₃ (δ 77.0 ppm) or C₆D₆ (δ 128.0 ppm) relative to Me₄Si at δ 0.00 ppm. The ³¹P chemical shifts were referenced to 85% H₃PO₄ (δ 0.00 ppm) as an external standard. Melting points are uncorrected. High-resolution mass spectra (HRMS) were measured using a Fourier transform mass spectrometer equipped with an electrospray ionization (ESI) or a double-focusing magnetic sector mass spectrometer equipped with electron impact (EI). All reagents were purchased and used without further purification. $K_3PO_4·H_2O$ was finely ground prior to use. THF and dioxane used as solvents were distilled from sodium/benzophenone ketyl prior to use. tert-Amyl alcohol used as solvent for Suzuki−Miyaura reactions was purchased and used as received. Solvents for extraction and chromatography were HPLC grade.

Preparation of CyR-Phos: [Boranatodi(cyclohexyl)phosphino]cyclopentadiene. Cyclopentadiene (0.62 mL, 7.54 mmol) was dissolved in THF (16 mL) and cooled to 0 °C. At this temperature a n-hexane solution of n-BuLi (4.5 mL, 1.59 M, 3.15 mmol) was added, and the reaction mixture was stirred for 1 h. To the resulting cyclopentadienyl anion was added HMPA (5.0 mL, 28.7 mmol), and the mixture was stirred for 10 min at 0 $^{\circ}$ C. Then, Cy₂PCl (530 μ L, 2.40 mmol) was added, and the reaction mixture was allowed to warm to room temperature. After being stirred for 3 h, the mixture was allowed to cool to -78 °C, and a THF solution of BH₃ (7.7 mL, 0.93 M, 7.16 mmol) was added and stirred for 30 min. The resulting mixture was allowed to warm to room temperature, quenched by 1 N HCl, diluted with $CHCl₃$ and washed twice with water and once with brine. The combined aqueous solutions were extracted with CHCl₃, and the combined organic solutions were dried over $Na₂SO₄$ and concentrated under reduced pressure. The residue was purified by silica gel chromatography $(1:1 \text{ CH}_2\text{Cl}_2/n\text{-}$ hexane) to yield [boranatodi(cyclohexyl)phosphino]cyclopentadiene (573.9 mg, 2.08 mmol, 87%, ca. 64:36 mixture of regioisomers based on ¹H NMR spectroscopic analysis) as a white solid: ¹H NMR (499.9 MHz, CDCl₃) δ 0−1 (m, 3H), 1.1−1.4 (m, 10H), 1.6−2.0 (m, 22H), 3.17 (s, 0.73H), 3.19 (s, 1.27H), 6.55−6.60 (m, 0.7H), 6.63−6.67 (m, 0.64H), 6.77−6.80 (m, 0.64H), 7.12−7.18 (m, 1H); 13C NMR (125.7 MHz, CDCl₃) δ 25.9 (d, J = 1.4 Hz), 26.0 (d, J = 1.4 Hz), 26.5, 26.61, 26.63, 26.65, 26.67, 26.69, 26.72, 26.76, 26.77, 26.82, 26.85, 26.86, 43.5 (d, J $= 12.2$ Hz), 45.3 (d, J = 7.3 Hz), 131.3 (d, J = 53.7 Hz), 132.3 (d, J = 50.5 Hz), 132.4 (d, J = 5.5 Hz), 132.6 (d, J = 12.7 Hz), 134.5 (d, J = 7.0 Hz), 139.8 (d, J = 4.8 Hz), 146.6 (d, J = 8.7 Hz), 148.5 (d, J = 10.2 Hz) (observed complexity due to P−C splitting); 31P NMR (202.3

MHz, CDCl₃) δ 21−22 (m); HRMS (EI) found 276.2178, calcd for $C_{17}H_{30}BP$ 276.2178.

CyR-Phos. [Boranatodi(cyclohexyl)phosphino]cyclopentadiene (313.8 mg, 1.14 mmol) was dissolved in THF (11 mL) and cooled to 0 \degree C. At this temperature a *n*-hexane solution of *n*-BuLi (0.68 mL, 1.63 M, 1.11 mmol) was added, and the reaction mixture was stirred for 1 h. To the resulting cyclopentadienyl anion was added a solution of the bromodicarbonyl ruthenium complex⁵ (153.3 mg, 0.225 mmol) in THF (10 mL) at 0 \degree C, and the reaction mixture was allowed to warm to room temperature and then stirred for an additional 18 h. The mixture was quenched by saturated aqueous $NH₄Cl$, diluted with $CHCl₃$, and washed twice with water and once with brine. The combined aqueous solutions were extracted with $CHCl₃$, and the combined organic solutions were dried over Na₂SO₄ and concentrated under reduced pressure. The residue was purified by silica gel chromatography (1:2 to 1:1 CHCl₃/n-hexane) to give the BH₃protected title compound (150.1 mg, 0.183 mol, 81%) as a pale-yellow solid, which was dissolved in a solution of DABCO (524.9 mg, 4.68 mmol) in dioxane (9 mL) and then heated to 100 °C. After being stirred for 1 h, the mixture was allowed to cool to room temperature and concentrated under reduced pressure. The residue was purified by silica gel chromatography (AcOEt) to give the title compound (146.9 mg, 0.182 mmol, 81% for 2 steps) as a pale-yellow solid: mp = 124− 126 °C; ¹H NMR (699.8 MHz, C₆D₆) δ 0.7−1.8 (m, 22H), 4.39 (s, 2H), 4.64 (s, 2H), 6.79 (ddt, $J = 7.7$ Hz, $J = 7.0$ Hz, $J = 1.4$ Hz, 1H), 6.85 (t, J = 7.7 Hz, 2H), 6.99–7.03 (m, 4H), 7.09 (dt, J = 7 Hz, J = 1.4 Hz, 2H), 7.14−7.16 (m, 2H), 7.19 (ddd, J = 8.4 Hz, J = 7.0 Hz, J = 1.4 Hz, 2H), 7.28−7.32 (m, 4H), 7.86 (dd, J = 8.4 Hz, J = 1.4 Hz, 2H), 8.15 (dd, J = 7.7 Hz, J = 0.7 Hz, 2H), 8.22 (d, J = 7.7 Hz, 2H); ¹³C NMR (176.0 MHz, C_6D_6) δ 26.8 (d, J = 0.9 Hz), 27.57 (d, J = 11.1 Hz), 27.63 (d, J = 7.9 Hz), 30.8 (d, J = 4.0 Hz), 30.9, 33.7 (d, J = 15.3 Hz), 76.7 (d, $J = 1.6$ Hz), 78.1 (d, $J = 10.7$ Hz), 84.4 (d, $J = 25.2$ Hz), 84.6, 93.3, 99.4, 124.0, 125.9, 126.4, 126.5, 127.0, 127.2, 127.4, 128.3, 128.4, 131.2, 132.3, 132.7, 133.1, 135.3, 135.4, 137.3 (one aromatic sp² carbon missing due to overlap); ³¹P NMR (283.3 MHz, C₆D₆) δ –9.8; HRMS (ESI) found 807.2685, calcd for $C_{52}H_{50}PRu [M + H]^+$ 807.2688.

Preparation of Aryl Chloride: 2,6-Diethylchlorobenzene $(1b).4c$ 2,6-Diethylaniline $(2.224 \text{ g}, 14.9 \text{ mmol})$ was dissolved in a mixture of HCl (12 N, 6.0 mL) and EtOH (16 mL) and cooled to 0 °C. [Th](#page-11-0)e mixture was added to an aqueous solution (6 mL) of NaNO₂ (1.413 g, 20.5 mmol) and then stirred for 30 min. The resulting diazonium salt solution was added to an aqueous HCl solution (12 N, 20 mL) of CuCl (789.2 mg, 8.0 mmol) and heated to reflux with vigorous stirring for 20 min. The mixture was allowed to cool to room temperature, quenched by NaHCO $_3$, extracted with *n*-hexane, and washed twice with water and once with brine. The organic solution was dried over $Na₂SO₄$ and concentrated under reduced pressure. The residue was purified by silica gel chromatography (n-hexane) to yield the title compound $(1.163 \text{ g}, 6.896 \text{ mmol}, 46\%)$ as a colorless oil. ¹H NMR (399.9 MHz, CDCl₃) δ 1.23 (t, J = 7.6 Hz, 6H), 2.78 (q, J = 7.6 Hz, 4H), 7.0−7.2 (m, 3H); ¹³C NMR (100.6 MHz, CDCl₃) δ 14.1, 27.3, 126.4, 126.9, 133.5, 141.9; MS (EI) m/z (%) 170 (M⁺ + 2, 21), 168 (M+ , 63), 155 (34), 153 (100), 133 (81); HRMS (EI) found 168.0706, calcd for $C_{10}H_{13}Cl_1$ 168.0706.

2,6-Diisopropylchlorobenzene (1c).^{4f} 2,6-Diisopropylaniline (4.307 g, 24.3 mmol) was dissolved in a mixture of HCl (12 N, 10 mL) and EtOH (25 mL) and cooled to 0 °[C.](#page-11-0) The mixture was added to an aqueous solution (10 mL) of NaNO_2 (2.299 g, 33.3 mmol) and then stirred for 60 min. The resulting diazonium salt solution was added to an aqueous HCl solution (12 N, 10 mL) of CuCl (1.345 g, 13.6 mmol) and heated to reflux with vigorous stirring for 50 min. The mixture was allowed to cool to room temperature, quenched by NaHCO₃, extracted with *n*-hexane, and washed twice with water and once with brine. The organic solution was dried over $\mathrm{Na}_2\mathrm{SO}_4$ and concentrated under reduced pressure. The residue was purified by silica gel chromatography $(n$ -hexane) to yield the title compound (2.339 g, 11.89 mmol, 49%) as a colorless oil. ¹ H NMR (399.9 MHz, CDCl₃) δ 1.24 (d, J = 6.8 Hz, 12H), 3.49 (sept, J = 6.8 Hz, 2H), 7.1− 7.2 (m, 3H); 13 C NMR (100.6 MHz, CDCl₃) δ 22.9, 30.6, 123.9,

126.6, 132.8, 146.0; MS (EI) m/z (%) 198 (M⁺+2, 9), 196 (M⁺, 26), 183 (33), 181 (100); HRMS (EI) found 196.1020, calcd for $C_{12}H_{17}Cl_1$ 196.1019.

2,4,6-Tri-tert-butylchlorobenzene $(1d).$ ¹³ 2,4,6-Tri-tert-butylaniline (1.496 g, 5.72 mmol) was dissolved in a mixture of HCl (12 N, 4.4 mL) and EtOH (10 mL) and cooled to [0](#page-11-0) °C. The mixture was added to an aqueous solution (4 mL) of NaNO₂ $(0.915 \text{ g}, 13.26)$ mmol) and then stirred for 60 min. The resulting diazonium salt solution was added to an aqueous HCl solution (12 N, 15 mL) of CuCl (0.526 g, 5.31 mmol) and heated to reflux with vigorous stirring for 60 min. The mixture was allowed to cool to room temperature, quenched by NaHCO₃, extracted with *n*-hexane, and washed twice with water and once with brine. The organic solution was dried over Na2SO4 and concentrated under reduced pressure. The residue was purified by silica gel chromatography (n-hexane) to yield the title compound (0.524 g, 1.87 mmol, 33%) as a white solid: mp = 157−158 $^{\circ}$ C; ¹H NMR (399.9 MHz, CDCl₃) δ 1.31 (s, 9H), 1.53 (s, 18H), 7.38 $(s, 2H);$ ¹³C NMR (100.6 MHz, CDCl₃) δ 30.5, 31.4, 34.9, 37.2, 123.1, 131.3, 147.0, 147.8; MS (EI) m/z (%) 282 (M⁺+2, 12), 280 (M⁺, 34), 267 (45), 265 (100); HRMS (EI) found 280.1960, calcd for $C_{18}H_{29}Cl_1$ 280.1958.

Representative Procedure for the Preparation of the Precatalyst Solution. A vial was charged with CyR-Phos (7.8 mg, 9.7 μ mol), sealed with a screw cap with a PTFE/silicone septum, and then evacuated and backfilled with argon (3 cycles). To the vial was sequentially added 200 μ L of a solution of Pd(dba)₂ (3.5 mg, 6.1) μ mol) in dioxane (0.5 mL) and ethyl 4-bromobenzoate (12.4 mg, 54 μ mol), and then the mixture was stored at room temperature for 3 h. The resulting solution was diluted with dioxane (1.8 mL) and then was used as the precatalyst solution.

General Procedure for Pd-Catalyzed Coupling of Aryl Halide with Arylboronic Acid. A flame-dried two-necked flask was charged with arylboronic acid and base, capped with a rubber septum, and then evacuated and backfilled with argon (3 cycles). To the flask was sequentially added aryl halide, the indicated solvent, and the indicated amount of a precatalyst solution via syringe through septum (aryl halides that were solids at room temperature were added during the initial charge, prior to the evacuation/backfill cycles), and then the reaction mixture was stirred at 100 °C for the indicated amount of time.

2,2',6-Trimethylbiphenyl (3aa)¹⁴ (Table 1, Entry 10). Following the general procedure, a mixture of 2,6-dimethylchlorobenzene (1a) (153.1 mg, 1.09 mmol), 2-meth[ylp](#page-11-0)henylboronic acid (2a) (210.6 mg, 1.55 mmol), K3PO4·H2O (719.8 mg, 3.[13](#page-1-0) mmol), a dioxane solution (88 μ L, 1.23 mM, 0.11 μ mol) of the precatalyst composed of $Pd(dba)₂$, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:4.1:21.6 in dioxane (2 mL) was stirred at 100 °C for 2 h. The resulting reaction mixture was diluted with n -hexane (5 mL), filtered through a thin pad of silica gel (n-hexane), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel (n-hexane) to provide the title compound (186.1 mg, 0.948 mmol, 87%) as a colorless oil. ¹H NMR (399.9 MHz, CDCl₃) δ 1.95 (s, 6H), 1.97 (s, 3H), 7.00–7.30 (m, 7H); ¹³C NMR (100.6 MHz, CDCl₃) δ 19.4, 20.3, 126.0, 126.9, 127.0, 127.2, 128.8, 129.9, 135.5, 135.8, 140.5, 141.0.

2,6-Dimethyl-2'-ethylbiphenyl $(3ab)^{5b}$ (Table 2, Entry 2). Following the general procedure, a mixture of 2,6-dimethylchlorobenzene (1a) (142.7 mg, 1.02 mmol), [2-e](#page-11-0)thylphenylboronic acid $(2b)$ (227.1 mg, 1.51 mmol), K₃PO₄·H₂O (696.9 mg, 3.[03](#page-1-0) mmol), and a dioxane solution (200 μ L, 1.22 mM, 0.24 μ mol) of the precatalyst composed of $Pd(dba)_2$, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:4.0:22.5 ratio in dioxane (2 mL) was stirred at 100 °C for 60 min. The resulting reaction mixture was diluted with *n*-hexane (5 mL) , filtered through a thin pad of silica gel (n-hexane), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel $(n$ -hexane) to provide the title compound (206.8 mg, 0.983 mmol, 96%) as a colorless oil. ¹H NMR (270.1 MHz, CDCl₃) δ 1.05 (t, J = 7.6 Hz, 3H), 1.96 (s, 6H), 2.28 (q, $J = 7.6$ Hz, 2H), 7.00 (d, $J = 6.8$ Hz, 1H), 7.05−7.40 (m, 6H); ¹³C NMR (100.6 MHz, CDCl₃) δ 14.5, 20.5, 25.7, 126.0, 126.9, 127.2,

128.2, 129.0, 136.0, 139.8, 140.9, 141.4 (one sp² carbon missing due to overlap); MS (EI) m/z (%) 210 (M⁺ , 77), 195 (100), 181 (38), 165 (50); HRMS (EI) found 210.1407, calcd for $C_{16}H_{18}$ 210.1409.

2,6-Dimethyl-2′-isopropylbiphenyl (3ac) (Table 2, Entry 3). Following the general procedure, a mixture of 2,6-dimethylchlorobenzene (1a) (140.0 mg, 1.00 mmol), 2-isopropylphenylboronic acid (2c) (243.9 mg, 1.49 mmol), $K_3PO_4·H_2O$ (704.3 mg, 3[.0](#page-1-0)6 mmol), a dioxane solution (200 μ L, 1.23 mM, 0.25 μ mol) of the precatalyst composed of $Pd(dba)_2$, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:4.1:21.6 ratio in dioxane (2 mL) was stirred at 100 $^{\circ}$ C for 80 min. The resulting reaction mixture was diluted with *n*-hexane (5 mL) , filtered through a thin pad of silica gel $(n$ -hexane), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel (n-hexane) to provide the title compound (171.4 mg, 0.764 mmol, 76%) as a colorless oil. ¹H NMR $(270.1 \text{ MHz}, \text{CDCl}_3) \delta 1.10 \text{ (d, } J = 6.8 \text{ Hz}, 6H), 1.97 \text{ (s, } 6H), 2.54$ (sept, J = 6.8 Hz, 1H), 6.98 (d, J = 7.3 Hz, 1H), 7.03−7.28 (m, 4H), 7.28−7.44 (m, 2H); 13C NMR (100.6 MHz, CDCl3) δ 20.7, 24.0, 29.8, 125.5, 125.8, 126.8, 127.2, 127.5, 129.0, 136.2, 139.0, 140.9, 146.3; MS (EI) m/z (%) 224 (M⁺, 76), 209 (100), 181 (48); HRMS (EI) found 224.1565, calcd for $C_{17}H_{20}$ 224.1565.

2,6-Dimethyl-2'-phenylbiphenyl (3ad)^{5b} (Table 2, Entry 4). Following the general procedure, a mixture of 2,6-dimethylchlorobenzene (1a) (140.3 mg, 1.00 mmol), 2-p[hen](#page-11-0)ylphenylboronic acid (2d) (297.4 mg, 1.50 mmol), $K_3PO_4·H_2O$ (698.4 mg, 3[.0](#page-1-0)3 mmol), a dioxane solution (200 μ L, 1.22 mM, 0.24 μ mol) of the precatalyst composed of $Pd(dba)₂$, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:3.9:22.5 ratio in dioxane (2 mL) was stirred at 100 °C for 80 min. The resulting reaction mixture was diluted with *n*-hexane (5 mL) , filtered through a thin pad of silica gel (n-hexane), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel $(n$ -hexane) to provide the title compound (258.5 mg, 1.00 mmol, 100%) as a colorless oil. ¹H NMR $(270.1 \text{ MHz}, \text{CDCl}_3) \delta 1.93 \text{ (s, 6H)}, 6.95 \text{ (d, } J = 7.3 \text{ Hz}, 2H), 7.00-$ 7.20 (m, 7H), 7.30–7.50 (m, 3H); ¹³C NMR (100.6 MHz, CDCl₃) δ 20.8, 126.5, 126.9, 127.1, 127.3, 127.4, 127.6, 128.7, 130.1, 130.3, 136.0, 138.9, 140.69, 140.74, 141.2; MS (EI) m/z (%) 258 (M⁺, 100), 243 (49), 228 (24); HRMS (EI) found 258.1409, calcd for $C_{20}H_{18}$ 258.1409.

2,6-Dimethyl-2'-methoxybiphenyl (3ae)^{5b} (Table 2, Entry 5). Following the general procedure, a mixture of 2,6-dimethylchlorobenzene (1a) (145.5 mg, 1.04 mmol), 2-met[hox](#page-11-0)yphenylboronic acid $(2e)$ (230.9 [m](#page-1-0)g, 1.52 mmol), K₃PO₄·H₂O (691.9 mg, 3.00 mmol), and a dioxane solution (200 μ L, 1.23 mM, 0.25 μ mol) of the precatalyst composed of $Pd(dba)₂$, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:4.1:21.6 ratio in dioxane (2 mL) was stirred at 100 °C for 120 min. The resulting reaction mixture was diluted with *n*-hexane (5 mL) , filtered through a thin pad of silica gel $(1:5 \text{ AcOEt}/n\text{-hexane})$, and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel (1:5 AcOEt/nhexane) to provide the title compound (193.6 mg, 0.912 mmol, 88%) as a colorless oil. ¹H NMR (270.1 MHz, CDCl₃) δ 2.01 (s, 6H), 3.73 $(s, 3H)$, 6.95−7.05 (m, 3H), 7.05−7.20 (m, 3H), 7.34 (m, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ 20.4, 55.2, 110.7, 120.6, 126.9, 128.3, 129.4, 130.5, 136.4, 138.1, 156.4 (one sp² carbon missing due to overlap); MS (EI) m/z (%) 212 (M⁺ , 100), 197 (36), 181 (46), 165 (32) ; HRMS (EI) found 212.1198, calcd for C₁₅H₁₆O 212.1201.

2,6-Dimethyl-2′-ethoxybiphenyl (3af) (Table 2, Entry 6). Following the general procedure, a mixture of 2,6-dimethylchlorobenzene (1a) (144.3 mg, 1.03 mmol), 2-ethoxyphenylboronic acid (2f) (249.0 mg, 1.50 mmol), K3PO4·H2O (695.6 mg, 3.[02](#page-1-0) mmol), and a dioxane solution (200 μ L, 1.23 mM, 0.25 μ mol) of the precatalyst composed of $Pd(dba)₂$, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:4.1:21.6 ratio in dioxane (2 mL) was stirred at 100 °C for 100 min. The resulting reaction mixture was diluted with *n*-hexane (5 mL) , filtered through a thin pad of silica gel (1:5 AcOEt/n-hexane), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel $(1:5 \text{ AcOE}t/n$ hexane) to provide the title compound (216.5 mg, 0.957 mmol, 93%) as a colorless oil. ¹H NMR (270.1 MHz, CDCl₃) δ 1.21 (t, J = 7.0 Hz,

3H), 2.02 (s, 6H), 3.99 (q, J = 7.0 Hz, 2H), 6.90−7.20 (m, 6H), 7.31 $(ddd, J = 8.1 \text{ Hz}, J = 6.8 \text{ Hz}, J = 2.7 \text{ Hz}, 1H$); ¹³C NMR (100.6 MHz, CDCl3) δ 14.6, 20.4, 63.4, 112.1, 120.5, 126.7, 126.8, 128.2, 129.9, 130.7, 136.4, 138.4, 155.7; MS (EI) m/z (%) 226 (M⁺, 100), 198 (38),

183 (83); HRMS (EI) found 226.1356, calcd for C₁₆H₁₈O 226.1358.
2,6-Dimethyl-2'-isopropoxybiphenyl (3ag)^{5b} (Table 2, Entry 7). Following the general procedure, a mixture of 2,6-dimethylchlorobenzene (1a) (141.0 mg, 1.00 mmol), 2-isopr[opo](#page-11-0)xyphenylboronic acid $(2g)$ $(310.3 \text{ mg}, 1.72 \text{ mmol})$ $(310.3 \text{ mg}, 1.72 \text{ mmol})$ $(310.3 \text{ mg}, 1.72 \text{ mmol})$, $K_3PO_4·H_2O$ $(707.7 \text{ mg}, 3.07$ mmol), and a dioxane solution (200 μ L, 1.22 mM, 0.24 μ mol) of the precatalyst composed of $Pd(dba)_{2}$, CyR-Phos, and ethyl 4bromobenzoate in a 1.0:4.0:22.5 ratio in dioxane (2 mL) was stirred at 100 °C for 80 min. The resulting reaction mixture was diluted with n-hexane (5 mL), filtered through a thin pad of silica gel (1:5 AcOEt/ n-hexane), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel (1:9 AcOEt/*n*-hexane) to provide the title compound (226.0 mg, 0.940 mmol, 94%) as a colorless oil. $^1\text{H NMR}$ (270.1 MHz, CDCl3) δ 1.12 (d, $J = 5.9$ Hz, 6H), 2.02 (s, 6H), 4.32 (sept, $J = 6.1$ Hz, 1H), 6.92−7.18 (m, 6H), 7.29 (td, J = 7.3 Hz, J = 2.2 Hz, 1H); 13C NMR $(100.6 \text{ MHz}, \text{CDCl}_3)$ δ 20.5, 22.0, 70.2, 114.7, 120.7, 126.7, 126.8, 128.1, 130.9, 131.1, 136.4, 138.6, 154.9; MS (EI) m/z (%) 240 (M⁺ , 61), 198 (99), 183 (100), 165 (25); HRMS (EI) found 240.1518, calcd for $C_{17}H_{20}O$ 240.1514.

2,6-Diethyl-2′-methylbiphenyl (3ba) (Table 3, Entry 1). Following the general procedure, a mixture of 2,6-diethylchlorobenzene (1b) (172.8 mg, 1.02 mmol), 2-methylphenylboronic acid (2a) (207.4 [m](#page-1-0)g, 1.53 mmol), $K_3PO_4·H_2O$ (722.2 mg, 3.14 mmol), and a dioxane solution (400 μ L, 1.25 mM, 0.50 μ mol) of the precatalyst composed of Pd(dba)₂, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:4.1:23.2 ratio in dioxane (2 mL) was stirred at 100 °C for 150 min. The resulting reaction mixture was diluted with *n*-hexane (5 mL) , filtered through a thin pad of silica gel $(n$ -hexane), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel $(n$ -hexane) to provide the title compound (212.8 mg, 0.949 mmol, 93%) as a colorless oil. $^1\rm H$ NMR $(399.9 \text{ MHz}, \text{CDCl}_3)$ δ 1.02 (t, J = 7.6 Hz, 6H), 1.96 (s, 3H), 2.17– 2.31 (m, 4H), 7.07 (d, J = 6.8 Hz, 1H), 7.16 (d, J = 7.6 Hz, 2H), 7.17− 7.30 (m, 4H); ¹³C NMR (100.6 MHz, CDCl₃) δ 15.1, 19.8, 26.3, 125.5, 127.0, 127.4, 129.70, 129.73, 135.9, 139.7, 139.8, 141.7 (one sp² carbon missing due to overlap); MS (EI) m/z (%) 224 (M⁺, 100), 209 (58), 195 (56), 181 (44), 165 (55); HRMS (EI) found 224.1567, calcd for $C_{17}H_{20}$ 224.1565.

2,2′,6-Triethylbiphenyl (3bb) (Table 3, Entry 2). Following the general procedure, a mixture of 2,6-diethylchlorobenzene (1b) (175.0 mg, 1.04 mmol), 2-ethylphenylboronic acid (2b) (228.0 mg, 1.52 mm[ol\)](#page-1-0), $K_3PO_4·H_2O$ (714.7 mg, 3.10 mmol), and a dioxane solution (400 μ L, 1.25 mM, 0.50 μ mol) of the precatalyst composed of $Pd(dba)₂$, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:4.1:23.2 ratio in dioxane (2 mL) was stirred at 100 °C for 90 min. The resulting reaction mixture was diluted with *n*-hexane (5 mL) , filtered through a thin pad of silica gel (n-hexane), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel (n-hexane) to provide the title compound (210.7 mg, 0.884 mmol, 85%) as a colorless oil. ¹H NMR (399.9 MHz, CDCl₃) δ 1.03 (t, J = 7.6 Hz, 6H), 1.05 (t, J = 7.6 Hz, 3H), 2.17−2.30 (m, 6H), 7.04 (d, J = 7.6 Hz, 1H), 7.15 (d, J = 7.6 Hz, 2H), 7.19−7.24 (m, 1H), 7.27−7.33 (m, 3H); 13C NMR (100.6 MHz, CDCl₃) δ 14.2, 15.1, 25.8, 26.6, 125.3, 125.4, 127.2, 127.4, 127.7, 129.8, 139.2, 139.6, 141.6, 141.9; MS (EI) m/z (%) 238 (M⁺ , 52), 209 (100), 181 (59), 165 (51); HRMS (EI) found 238.1723, calcd for $C_{18}H_{22}$ 238.1722.

2,6-Diethyl-2′-isopropylbiphenyl (3bc) (Table 3, Entry 3). Following the general procedure, a mixture of 2,6-diethylchlorobenzene (1b) (163.3 mg, 0.97 mmol), 2-isopropylphenylboronic acid (2c) (249.4 [m](#page-1-0)g, 1.52 mmol), $K_3PO_4·H_2O$ (696.1 mg, 3.02 mmol), and a dioxane solution (400 μ L, 1.30 mM, 0.52 μ mol) of the precatalyst composed of $Pd(dba)₂$, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:3.8:17.3 ratio in dioxane (2 mL) was stirred at 100 $^{\circ}{\rm C}$ for 90 min. The resulting reaction mixture was diluted with *n*-hexane (5 mL) ,

filtered through a thin pad of silica gel $(n$ -hexane), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel (n-hexane) to provide the title compound (229.8 mg, 0.910 mmol, 94%) as a colorless oil. ¹H NMR $(399.9 \text{ MHz}, \text{CDCl}_3)$ δ 1.06 (t, J = 7.6 Hz, 6H), 1.08 (d, J = 6.8 Hz, 6H), 2.16−2.33 (m, 4H), 2.53 (sept, J = 6.8 Hz, 1H), 7.02 (ddd, J = 7.6 Hz, $J = 1.2$ Hz, $J = 0.4$ Hz, 1H), 7.16 (d, $J = 7.6$ Hz, 2H), 7.20 $(ddd, J = 7.6 \text{ Hz}, J = 6.8 \text{ Hz}, J = 1.6 \text{ Hz}, 1H), 7.29 \text{ (dd, } J = 8.0 \text{ Hz}, J =$ 6.8 Hz, 1H), 7.32-7.41 (m, 2H); ¹³C NMR (100.6 MHz, CDCl₃) δ 15.2, 23.9, 26.6, 29.8, 125.20, 125.26, 125.3, 127.3, 127.5, 129.8, 138.4, 139.6, 142.1, 146.5; MS (EI) m/z (%) 252 (M⁺, 87), 223 (59), 209 (69), 181 (100); HRMS (EI) found 252.1876, calcd for $C_{19}H_{24}$ 252.1878.

2,6-Diethyl-2′-phenylbiphenyl (3bd) (Table 3, Entry 4). Following the general procedure, a mixture of 2,6-diethylchlorobenzene (1b) (168.2 mg, 1.00 mmol), 2-phenylphenylboronic acid (2d) $(301.1 \text{ mg}, 1.52 \text{ mmol})$ $(301.1 \text{ mg}, 1.52 \text{ mmol})$ $(301.1 \text{ mg}, 1.52 \text{ mmol})$, $K_3PO_4·H_2O$ $(691.0 \text{ mg}, 3.00 \text{ mmol})$, and a dioxane solution (400 μ L, 1.25 mM, 0.50 μ mol) of the precatalyst composed of $Pd(dba)₂$, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:4.1:23.2 ratio in dioxane (2 mL) was stirred at 100 °C for 170 min. The resulting reaction mixture was diluted with *n*-hexane (5 mL) , filtered through a thin pad of silica gel $(n$ -hexane), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel $(n$ -hexane) to provide the title compound (266.7 mg, 0.931 mmol, 93%) as a white solid: mp = 61− 62 °C; ¹H NMR (399.9 MHz, CDCl₃) δ 0.98 (t, J = 7.6 Hz, 6H), 2.16−2.32 (m, 4H), 7.02 (d, J = 7.6 Hz, 2H), 7.05−7.08 (m, 2H), 7.10−7.17 (m, 3H), 7.17−7.21 (m, 2H), 7.37 (td, J = 7.2 Hz, J = 1.6 H, 1), 7.42−7.49 (m, 2H); 13C NMR (100.6 MHz, CDCl3) δ 14.8, 26.5, 125.1, 126.5, 126.9, 127.3, 127.4, 127.5, 128.9, 123.0, 130.9, 138.4, 139.5, 140.8, 141.2, 141.9; MS (EI) m/z (%) 286 (M⁺, 100), 271 (15), 257 (49), 241 (22), 229 (21), 165 (16); HRMS (EI) found 286.1723, calcd for C₂₂H₂₂ 286.1722.

2,6-Diethyl-2′-methoxybiphenyl (3be) (Table 3, Entry 5). Following the general procedure, a mixture of 2,6-diethylchlorobenzene (1b) (175.1 mg, 1.04 mmol), 2-methoxyphenylboronic acid (2e) $(229.4 \text{ mg}, 1.51 \text{ mmol})$, $K_3PO_4 \cdot H_2O$ (695.1 mg, 3.[02](#page-1-0) mmol), a dioxane solution (400 μ L, 1.30 mM, 0.52 μ mol) of the precatalyst composed of $Pd(dba)₂$, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:3.8:17.3 ratio in dioxane (2 mL) was stirred at 100 °C for 150 min. The resulting reaction mixture was diluted with *n*-hexane (5 mL) , filtered through a thin pad of silica gel (AcOEt), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel $(0:1$ to $1:0$ AcOEt/n-hexane) to provide the title compound (227.4 mg, 0.946 mmol, 91%) as a colorless oil. ¹H NMR (399.9 MHz, CDCl₃) δ 1.01 (t, J = 7.6 Hz, 6H), 2.22−2.38 (m, 4H), 3.72 (s, 3H), 6.96−7.02 (m, 2H), 7.05 (dd, J = 7.2 Hz, $J = 1.6$ Hz, 1H), 7.15 (d, $J = 8.0$ Hz, 2H), 7.28 (dd, $J = 8.4$ Hz, $J =$ 7.2 Hz, 1H), 7.35 (ddd, J = 8.0 Hz, J = 7.2 Hz, J = 2.0 Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ 15.0, 26.7, 55.0, 110.4, 120.2, 125.3, 127.5, 128.3, 128.8, 131.1, 136.9, 142.5, 156.8; MS (EI) m/z (%) 240 (M⁺ , 100), 225 (18), 211 (74), 196 (25), 181 (33), 165 (39), 152 (21); HRMS (EI) found 240.1513, calcd for $C_{17}H_{20}O$ 240.1514.

2,6-Diethyl-2′-ethoxybiphenyl (3bf) (Table 3, Entry 6). Following the general procedure, a mixture of 2,6-diethylchlorobenzene (1b) (169.6 mg, 1.01 mmol), 2-ethoxyphenylboronic acid (2f) (250.7 [m](#page-1-0)g, 1.51 mmol), $K_3PO_4·H_2O$ (696.2 mg, 3.02 mmol), and a dioxane solution (400 μ L, 1.30 mM, 0.52 μ mol) of the precatalyst composed of $Pd(dba)₂$, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:3.8:17.3 ratio in dioxane (2 mL) was stirred at 100 °C for 150 min. The resulting reaction mixture was diluted with *n*-hexane (5 mL) , filtered through a thin pad of silica gel $(1:5 \text{ AcOE}t/n$ -hexane), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel $(1:9$ AcOEt/nhexane) to provide the title compound (243.7 mg, 0.958 mmol, 95%) as a colorless oil. ¹H NMR (399.9 MHz, CDCl₃) δ 1.01 (t, J = 7.6 Hz, 6H), 1.19 (t, J = 6.8 Hz, 3H), 2.22–2.38 (m, 4H), 3.98 (q, J = 6.8 Hz, 2H), 6.92−6.99 (m, 2H), 7.04 (dd, J = 7.6 Hz, J = 2.0 Hz, 1H), 7.13 $(d, J = 7.6 \text{ Hz}, 2H), 7.25 \text{ (t, } J = 7.6 \text{ Hz}, 1H), 7.31 \text{ (ddd, } J = 8.4 \text{ Hz}, J =$ 7.6 Hz, J = 2 Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ 14.5, 15.1,

26.7, 63.1, 111.5, 120.0, 125.1, 127.1, 128.2, 129.3, 131.3, 137.2, 142.5, 156.1; MS (EI) m/z (%) 254 (M⁺, 100), 225 (100), 197 (64), 181 (26), 165 (21); HRMS (EI) found 254.1669, calcd for $C_{18}H_{22}O$ 254.1671.

2,6-Diethyl-2′-isopropoxybiphenyl (3bg) (Table 3, Entry 7). Following the general procedure, a mixture of 2,6-diethylchlorobenzene (1b) (167.7 mg, 0.99 mmol), 2-isopropoxyphenylboronic acid $(2g)$ $(2g)$ $(2g)$ (272.1 [m](#page-1-0)g, 1.51 mmol), K₃PO₄·H₂O (695.6 mg, 3.02 mmol), and a dioxane solution (400 μ L, 1.30 mM, 0.52 μ mol) of the precatalyst composed of $Pd(dba)₂$, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:3.8:17.3 ratio in dioxane (2 mL) was stirred at 100 °C for 210 min. The resulting reaction mixture was diluted with *n*-hexane (5 mL) , filtered through a thin pad of silica gel (AcOEt), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel $(0:1$ to $1:0$ AcOEt/n-hexane) to provide 251.7 mg of a mixture of the title compound and isopropoxybenzene in a 89.3:10.7 ratio as a colorless oil, which corresponds to a 90% yield of the title compound (237.5 mg, 0.889 mmol). The title compound: ¹H NMR (399.9 MHz, CDCl₃) δ 1.02 (t, J = 7.6 Hz, 6H), 1.13 (d, J = 6.0 Hz, 6H), 2.22−2.40 (m, 4H), 4.40 $(s$ ept, J = 6.0 Hz, 1H), 6.92–6.98 (m, 2H), 7.05 (dd, J = 7.2 Hz, J = 1.6 Hz, 1H), 7.12 (d, J = 7.6 Hz, 2H), 7.22−7.32 (m, 2H); 13C NMR $(100.6 \text{ MHz}, \text{CDCl}_3) \delta 15.1, 21.9, 26.7, 69.4, 113.1, 120.0, 125.1,$ 127.1, 128.0, 130.1, 131.5, 137.4, 142.4, 155.2; MS (EI) m/z (%) 268 (M+ , 45), 226 (73), 197 (100); HRMS (EI) found 268.1828, calcd for $C_{19}H_{24}O$ 268.1827.

2,6-Diisopropyl-2′-methylbiphenyl (3ca) (Table 4, Entry 1). Following the general procedure, a mixture of 2,6-diisopropylchlorobenzene (1c) (201.8 mg, 1.03 mmol), 2-methylphenylboronic acid $(2a)$ (207.4 mg, [1](#page-2-0).53 [m](#page-2-0)mol), $K_3PO_4·H_2O$ (692.1 mg, 3.01 mmol), and a dioxane solution (400 μ L, 1.25 mM, 0.50 μ mol) of the precatalyst composed of $Pd(dba)₂$, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:4.0:21.2 ratio in dioxane (2 mL) was stirred at 100 °C for 240 min. The resulting reaction mixture was diluted with *n*-hexane (5 mL) , filtered through a thin pad of silica gel $(n$ -hexane), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel $(n$ -hexane) to provide the title compound (165.1 mg, 0.654 mmol, 64%) as a colorless oil, 2,2′ dimethylbiphenyl (38.1 mg, 0.209 mmol, 27%), and 1,3-diisopropylbenzene (23.6 mg, 0.145 mmol, 14%). The title compound: ¹H NMR $(399.9 \text{ MHz}, \text{CDCl}_3)$ δ 1.02 (d, J = 6.8 Hz, 6H), 1.11 (d, J = 7.2 Hz, 6H), 1.99 (s, 3H), 2.45 (sept, J = 6.8 Hz, 2H), 7.05 (d, J = 6.8 Hz, 1H), 7.18−7.32 (m, 3H), 7.22 (d, J = 8.0 Hz, 2H), 7.35 (dd, J = 7.6, J $= 7.2$ Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ 20.2, 23.4, 24.9, 30.3, 122.7, 125.4, 126.9, 127.7, 129.6, 129.8, 136.3, 138.4, 139.9, 146.5; MS (EI) m/z (%) 252 (M⁺, 38), 237 (24), 195 (100), 179 (25), 167 (48), 165 (38); HRMS (EI) found 252.1878, calcd for $C_{19}H_{24}$ 252.1878.

2,6-Diisopropyl-2′-ethylbiphenyl (3cb) (Table 4, Entry 2). Following the general procedure, a mixture of 2,6-diisopropylchlorobenzene (1c) (204.8 mg, 1.04 mmol), 2-ethylphenylboronic acid (2b) (226.1 mg, 1.51 mmol), K₃PO₄·H₂O (690.9 mg, 3.[00](#page-2-0) mmol), and a dioxane solution (400 μ L, 1.20 mM, 0.48 μ mol) of the precatalyst composed of $Pd(dba)_{2}$, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:4.1:25.8 ratio in dioxane (2 mL) was stirred at 100 °C for 240 min. The resulting reaction mixture was diluted with *n*-hexane (5 mL) , filtered through a thin pad of silica gel $(n$ -hexane), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel (n-hexane) to provide the title compound (86.6 mg, 0.325 mmol, 31%) as a colorless oil, 2,2′ diethylbiphenyl (34.5 mg, 0.164 mmol, 22%), and 1,3-diisopropylbenzene (13.1 mg, 0.081 mmol, 8%). The title compound: ¹H NMR $(399.9 \text{ MHz}, \text{CDCl}_3)$ δ 1.01 (d, J = 6.8 Hz, 6H), 1.10 (t, J = 7.6 Hz, 6H), 1.12 (d, $J = 6.8$ Hz, 6H), 2.30 (q, $J = 7.6$ Hz, 2H), 2.46 (sept, $J =$ 7.2 Hz, 2H), 7.03 (d, J = 7.2 Hz, 1H), 7.20−7.25 (m, 3H), 7.30−7.40 (m, 3H); 13C NMR (100.6 MHz, CDCl3) δ 14.1, 23.2, 25.1, 25.7, 30.3, 122.6, 125.2, 127.1, 127.5, 127.7, 129.9, 138.2, 139.2, 141.9, 146.7; MS (EI) m/z (%) 266 (M+ , 46), 209 (58), 181 (100); HRMS (EI) found 266.2032, calcd for $C_{20}H_{26}$ 266.2035.

2,2′,6-Trisopropylbiphenyl (3cc) (Table 4, Entry 3). Following the general procedure, a mixture of 2,6-diisopropylchlorobenzene (1c) (197.5 mg, 1.00 mmol), 2-isopropylphenylboronic acid (2c) (249.7 mg, 1.52 [m](#page-2-0)mol), $K_3PO_4·H_2O$ (693.3 mg, 3.01 mmol), and a dioxane solution (400 μ L, 1.20 mM, 0.48 μ mol) of the precatalyst composed of $Pd(dba)₂$, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:4.1:25.8 ratio in dioxane (2 mL) was stirred at 100 °C for 270 min. The resulting the reaction mixture was diluted with *n*-hexane (5 mL) , filtered through a thin pad of silica gel (n-hexane), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel (n-hexane) to provide the title compound (75.0 mg, 0.267 mmol, 27%) as a white solid, 2,2′ diisopropylbiphenyl (25.4 mg, 0.107 mmol, 14%) as a colorless oil, and 1,3-diisopropylbenzene (19.2 mg, 0.118 mmol, 12%). The title compound: mp = 59–60 °C; ¹H NMR (399.9 MHz, CDCl₃) δ 1.00 $(d, J = 6.8 \text{ Hz}, 6\text{H}), 1.10 (d, J = 6.8 \text{ Hz}, 6\text{H}), 1.15 (d, J = 7.2 \text{ Hz}, 6\text{H}),$ 2.48 (sept, J = 6.8 Hz, 2H), 2.62 (sept, J = 7.2 Hz, 1H), 7.01 (d, J = 7.2 Hz, 1H), 7.15−7.25 (m, 3H), 7.30−7.45 (m, 3H); 13C NMR (100.6 MHz, CDCl3) δ 23.0, 23.9, 25.4, 29.6, 30.3, 122.6, 125.1, 125.3, 127.4, 127.6, 129.9, 138.1, 138.3, 146.9, 147.0; MS (EI) m/z (%) 280 (M⁺ , 28) 237 (18), 223 (16), 195 (100), 181 (53); HRMS (EI) found 280.2186, calcd for $C_{21}H_{28}$ 280.2191.

2,6-Diisopropyl-2′-phenylbiphenyl (3cd) (Table 4, Entry 4). Following the general procedure, a mixture of 2,6-diisopropylchlorobenzene (1c) (193.6 mg, 0.98 mmol), 2-phenylphenylboronic acid $(2d)$ (297.2 mg, [1](#page-2-0).50 [m](#page-2-0)mol), K₃PO₄·H₂O (693.9 mg, 3.01 mmol), and a dioxane solution (400 μ L, 1.20 mM, 0.48 μ mol) of the precatalyst composed of $Pd(dba)₂$, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:4.1:25.8 ratio in dioxane (2 mL) was stirred at 100 °C for 17.5 h. The resulting the reaction mixture was diluted with *n*-hexane (5 mL) , filtered through a thin pad of silica gel $(n$ -hexane), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel $(n$ -hexane) to provide the title compound (117.1 mg, 0.372 mmol, 38%) as a white solid, 2,2′ diphenylbiphenyl (89.1 mg, 0.291 mmol, 39%) as a white solid, and 1,3-diisopropylbenzene (17.5 mg, 0.108 mmol, 11%). The title compound: mp = 129–133 °C; ¹H NMR (399.9 MHz, CDCl₃) δ 0.86 (d, J = 6.8 Hz, 6H), 1.01 (d,, J = 6.8 Hz, 6H), 2.56 (sept, J = 6.8 Hz, 2H), 7.05−7.28 (m, 9H), 7.37 (td, J = 7.2 Hz, J = 1.2 Hz, 1H), 7.44 (td, $J = 7.2$ Hz, $J = 1.2$ Hz, 1H), 7.49 (dd, $J = 7.6$ Hz, $J = 1.6$ Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ 22.7, 25.4, 30.4, 122.5, 126.5, 126.7, 127.4, 127.5, 127.8, 129.3, 129.7, 131.0, 138.1, 138.4, 141.00, 141.01, 146.6; MS (EI) m/z (%) 314 (M⁺, 84), 271 (82), 257 (100); HRMS (EI) found 314.2032, calcd for $C_{24}H_{26}$ 314.2035.

2,6-Diisopropyl-2′-methoxybiphenyl (3ce) (Table 4, Entry 5). Following the general procedure, a mixture of 2,6-diisopropylchlorobenzene (1c) (193.6 mg, 0.984 mmol), 2-methoxyphenylboronic acid $(2e)$ (231.0 [m](#page-2-0)g, 1.52 mmol), K₃PO₄·H₂O (699.7 mg, 3.04 mmol), and a dioxane solution (400 μ L, 1.20 mM, 0.48 μ mol) of the precatalyst composed of $Pd(dba)₂$, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:4.1:25.8 ratio in dioxane (2 mL) was stirred at 100 °C for 180 min. The resulting reaction mixture was diluted with *n*-hexane (5 mL) , filtered through a thin pad of silica gel (AcOEt), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel (n-hexane) to provide the title compound (218.3 mg, 0.813 mmol, 83%) as a white solid: mp = 50− 51 °C; ¹H NMR (399.9 MHz, CDCl₃) δ 1.06 (d, J = 6.8 Hz, 6H), 1.08 $(d, J = 6.8 \text{ Hz}, 6\text{H})$, 2.55 (sept, $J = 6.8 \text{ Hz}, 2\text{H}$), 3.71 (s, 3H), 6.96− 7.06 (m, 3H), 7.21 (d, J = 7.6 Hz, 2H), 7.32−7.39 (m, 2H); 13C NMR (100.6 MHz, CDCl3) δ 23.7, 24.2, 30.5, 55.0, 110.3, 120.2, 122.4, 127.8, 128.3, 129.0, 131.2, 135.6, 147.1, 157.1; MS (EI) m/z (%) 268 (M⁺ , 59), 225 (100), 211 (73); HRMS (EI) found 268.1827, calcd for $C_{19}H_{24}O$ 268.1827.

2,6-Diisopropyl-2′-ethoxybiphenyl (3cf) (Table 4, Entry 6). Following the general procedure, a mixture of 2,6-diisopropylchlorobenzene (1c) (191.2 mg, 0.972 mmol), 2-ethoxyphenylboronic acid $(2f)$ (25[1](#page-2-0).4 mg, 1.51 mmol), K₃PO₄·H₂O (692.6 mg, 3.01 mmol), and a dioxane solution (400 μ L, 1.20 mM, 0.48 μ mol) of the precatalyst composed of $Pd(dba)₂$, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:4.1:25.8 ratio in dioxane (2 mL) was stirred at 100 °C for 330 min.

The resulting reaction mixture was diluted with *n*-hexane (5 mL) , filtered through a thin pad of silica gel (AcOEt), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel (n-hexane) to provide the title compound (229.4 mg, 0.812 mmol, 84%) as a colorless oil. ¹H NMR $(399.9 \text{ MHz}, \text{CDCl}_3)$ δ 1.06 (d, J = 6.8 Hz, 6H), 1.08 (d, J = 6.8 Hz, 6H), 1.20 (t, $J = 6.8$ Hz, 3H), 2.56 (sept, $J = 6.8$ Hz, 2H), 3.99 (q, $J =$ 6.8 Hz, 2H), 6.92–7.00 (m, 2H), 7.03 (dd, J = 7.6 Hz, J = 2.0 Hz, 1H), 7.20 (d, J = 7.6 Hz, 2H), 7.30−7.36 (m, 2H); 13C NMR (100.6 MHz, CDCl3) δ 14.6, 23.8, 24.2, 30.5, 62.9, 111.1, 119.9, 122.2, 127.6, 128.1, 129.3, 131.3, 135.9, 147.1, 156.3; MS (EI) m/z (%) 282 (M⁺, 78), 239 (70), 225 (49), 211 (34), 197 (100); HRMS (EI) found 282.1982, calcd for $C_{20}H_{26}O$ 282.1984.

2,6-Diisopropyl-2′-isopropoxybiphenyl (3cg) (Table 4, Entry 7). Following the general procedure, a mixture of 2,6-diisopropylchlorobenzene (1c) (193.8 mg, 0.985 mmol), 2-isopropoxyphenylboronic acid $(2g)$ $(278.0 \text{ mg}, 1.54 \text{ mmol})$ $(278.0 \text{ mg}, 1.54 \text{ mmol})$ $(278.0 \text{ mg}, 1.54 \text{ mmol})$, $K_3PO_4H_2O$ $(716.6 \text{ mg}, 3.11)$ mmol), and a dioxane solution (400 μ L, 1.20 mM, 0.48 μ mol) of the precatalyst composed of $Pd(dba)_2$, CyR-Phos, and ethyl 4bromobenzoate in a 1.0:4.1:25.8 ratio in dioxane (2 mL) was stirred at 100 °C for 240 min. The resulting reaction mixture was diluted with *n*-hexane (5 mL) , filtered through a thin pad of silica gel $(n$ -hexane), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel (1:30 AcOEt/n-hexane) to provide the title compound (200.4 mg, 0.676 mmol, 69%) as a white solid, 2,2′-diisopropoxybiphenyl (26.0 mg, 0.096 mmol, 13%) as a colorless oil, and 1,3-diisopropylbenzene (8.0 mg, 0. 049 mmol, 5%). The title compound: mp = 61.1–65.9 °C; ¹H NMR (399.9 MHz, CDCl₃) δ 1.04 (d, J = 6.8 Hz, 6H), 1.09 (d, J = 6.8 Hz, 6H), 1.15 (d, $J = 6.0$ Hz, 6H), 2.56 (sept, $J = 6.8$ Hz, 2H), 4.46 (sept, J = 6.0 Hz, 1H), 6.90−6.98 (m, 2H), 7.02 (dd, J = 7.2 Hz, J = 1.6 Hz, 1H), 7.18 (d, J = 8.0 Hz, 2H), 7.28–7.38 (m, 2H); ¹³C NMR $(100.6 \text{ MHz}, \text{CDCl}_3)$ δ 21.9, 23.8, 24.4, 30.4, 68.8, 112.5, 119.8, 122.1, 127.4, 128.0, 130.0, 131.6, 136.1, 147.1, 155.2; MS (EI) m/z (%) 296 (M+ , 46), 254 (71), 239 (26), 211 (97), 197 (100), 169 (47); HRMS (EI) found 296.2140, calcd for $C_{21}H_{28}O$ 296.2140.

1-Methoxy-4-[2-methyl-2-(3,5-di-tert-butylphenyl)propyl] benzene (4eh) (Table 5, Entry 2). Following the general procedure, a mixture of 2,4,6-tri-tert-butylbromobenzene (1e) (325.4 mg, 1.00 mmol), 4-methoxyphenylboronic acid (2h) (230.4 mg, 1.52 mmol), K₃PO₄·H₂O (69[3.](#page-2-0)3 mg, 3.01 mmol), and a dioxane solution (200 μ L, 1.30 mM, 0.26 μ mol) of the precatalyst composed of Pd(dba)₂, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:3.8:22.7 ratio in dioxane (2 mL) was stirred at 100 °C for 9 h. The resulting reaction mixture was diluted with n-hexane (5 mL), filtered through a thin pad of silica gel (1:5 to 1:0 AcOEt/n-hexane), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel $(1:2 \text{ CHCl}_3/n\text{-}$ hexane) to provide the title compound $(331.6 \text{ mg}, 0.940 \text{ mmol}, 94\%)$ as a white solid: mp = 81−82 °C; ¹ H NMR (699.8 MHz, CDCl3) δ 1.30 (s, 18H), 1.31 (s, 6H), 2.76 (s, 2H), 3.74 (s, 3H), 6.67 (s, 4H), 7.10 (d, $J = 1.4$ Hz, 2H), 7.25 (t, J = 1.8 Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ 28.1, 31.6, 34.9, 39.1, 50.7, 55.0, 112.8, 119.3, 120.5, 131.1, 131.2, 147.7, 149.7, 157.8; MS (EI) m/z (%) 352 (M⁺, 4), 231 (100); HRMS (EI) found 352.2765, calcd for C₂₅H₃₆O 352.2766.

1-Methoxy-2-[2-methyl-2-(3,5-di-t*ert*-butylphenyl)propyl]-
benzene (4ea)^{10a} (Table 5, Entry 3). Following the general procedure, a mixture of 2,4,6-tri-tert-butylbromobenzene (1e) (325.3 mg, 1.00 mmol)[, 2-](#page-11-0)methylphenylboronic acid (2a) (202.9 mg, 1.49 mmol), $K_3PO_4·H_2O$ (689.2 [mg,](#page-2-0) 2.99 mmol), and a dioxane solution (460 μ L, 1.11 mM, 0.51 μ mol) of the precatalyst composed of $Pd(dba)_{2}$, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:4.6:22.3 ratio in dioxane (2 mL) was stirred at 100 °C for 8 h. The resulting reaction mixture was diluted with n -hexane (5 mL), filtered through a thin pad of silica gel (n-hexane), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel (n-hexane) to provide the title compound (336.1 mg, 0.999 mmol, 100%) as a colorless oil. ¹H NMR (399.9 MHz, CDCl₃) δ 1.25 (s, 18H), 1.41 (s, 6H), 1.73 (s, 3H), 2.82 (s, 2H), 6.82 (dd, J = 7.2 Hz, J = 1.2 Hz, 1H), 6.95−7.07

 $(m, 3H)$, 7.01 (d, J = 2.0 Hz, 2H), 7.24 (t, J = 2.0 Hz, 1H); ¹³C NMR (176.0 MHz, CDCl3) δ 19.4, 28.5, 34.9, 39.7, 47.5, 119.5, 120.5, 124.7, 125.8, 130.1, 131.3, 137.3, 137.6, 147.2, 149.7; MS (EI) m/z (%) 336 $(M^{+}, 3.1)$, 231 (100); HRMS (EI) found 336.2817, calcd for $C_{25}H_{36}$ 336.2817.

1-Ethoxy-2-[2-methyl-2-(3,5-di-tert-butylphenyl)propyl] benzene (4eb) (Table 5, Entry 4). Following the general procedure, a mixture of 2,4,6-tri-tert-butylbromobenzene (1e) (325.7 mg, 1.00 mmol), 2-ethylphenylboronic acid (2b) (225.4 mg, 1.50 mmol), K₃PO₄·H₂O (717.1 mg, [3.](#page-2-0)11 mmol), and a dioxane solution (480 μ L, 1.11 mM, 0.53 μ mol) of the precatalyst composed of Pd(dba)₂, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:5.0:23.3 ratio in dioxane (2 mL) was stirred at 100 °C for 8 h. The resulting reaction mixture was diluted with n-hexane (5 mL), filtered through a thin pad of silica gel (n-hexane), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel (n-hexane) to provide the title compound (319.7 mg, 0.912 mmol, 91%) as a colorless oil. ¹H NMR (399.9 MHz, CDCl₃) δ 0.97 (t, J = 7.6 Hz, 3H), 1.26 (s, 18H), 1.40 (s, 6H), 2.09 (q, J = 7.6 Hz, 2H), 2.84 $(s, 2H)$, 6.80 (dd, J = 7.6 Hz, J = 1.2 Hz, 1H), 6.96–7.05 (m, 2H), 7.01 (d, J = 2.0 Hz, 2H), 7.01 (td, J = 7.6 Hz, J = 1.2 Hz, 1H), 7.24 (t, J $= 2.0$ Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ 15.1, 25.1, 28.5, 31.5, 34.9, 39.6, 46.7, 119.5, 120.5, 124.6, 126.1, 128.0, 131.4, 136.5, 143.5, 147.4, 149.7; MS (EI) m/z (%) 350 (M⁺, 1.8), 231 (100); HRMS (EI) found 350.2973, calcd for $C_{26}H_{38}$ 350.2974.

1-Isopropyl-2-[2-methyl-2-(3,5-di-tert-butylphenyl)propyl] benzene (4ec) (Table 5, Entry 5). Following the general procedure, a mixture of 2,4,6-tri-tert-butylbromobenzene (1e) (326.0 mg, 1.00 mmol), 2-isopropylphenylboronic acid (2c) (246.7 mg, 1.50 mmol), $K_3PO_4·H_2O$ (689.8 mg, [3](#page-2-0).00 mmol), and a dioxane solution (460 μ L, 1.11 mM, 0.51 μ mol) of the precatalyst composed of Pd(dba)₂, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:4.6:22.3 ratio in dioxane (2 mL) was stirred at 100 °C for 8 h. The resulting reaction mixture was diluted with n-hexane (5 mL), filtered through a thin pad of silica gel (n-hexane), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel $(n$ -hexane) to provide the title compound $(299.5 \text{ mg}, 0.821 \text{ mmol})$ 82%) as a colorless oil. ¹H NMR (399.9 MHz, CDCl₃) δ 0.94 (d, J = 6.8 Hz, 3H), 1.26 (s, 18H), 1.40 (s, 6H), 2.79 (sept, J = 6.8 Hz, 1H), 2.91 (s, 2H), 6.81 (d, J = 7.2 Hz, 1H), 6.95−7.00 (m, 1H), 7.05 (d, J = 2.0 Hz, 2H), 7.10−7.15 (m, 2H), 7.23 (t, J = 2.0 Hz, 1H); 13C NMR $(100.6 \text{ MHz}, \text{CDCl}_3)$ δ 23.7, 28.7, 28.8, 31.9, 34.9, 39.5, 46.1, 119.6, 120.6, 124.4, 124.7, 126.2, 131.4, 135.7, 147.5, 147.8, 149.6; MS (EI) m/z (%) 364 (M+ , 1.2), 231 (100); HRMS (EI) found 364.3131, calcd for $C_{27}H_{40}$ 364.3130.

1-Phenyl-2-[2-methyl-2-(3,5-di-*tert*-butylphenyl)propyl]-
benzene (4ed)^{10a} (Table 5, Entry 6). Following the general procedure, a mixture of 2,4,6-tri-tert-butylbromobenzene (1e) (324.8 mg, 1.00 mmol)[, 2-](#page-11-0)phenylphenylboronic acid (2d) (295.8 mg, 1.49 mmol), $K_3PO_4·H_2O$ (697.1 [mg,](#page-2-0) 3.03 mmol), and a dioxane solution (480 μ L, 1.11 mM, 0.53 μ mol) of the precatalyst composed of $Pd(dba)₂, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:4.7:22.3$ ratio in dioxane (2 mL) was stirred at 100 °C for 12 h. The resulting reaction mixture was diluted with n -hexane (5 mL) , filtered through a thin pad of silica gel (n-hexane), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel (n-hexane) to provide the title compound (379.6 mg, 0.952 mmol, 95%) as a white solid: mp = 55−56 °C; ¹ H NMR (699.8 MHz, CDCl3) δ 1.07 (s, 6H), 1.26 (s, 18H), 2.99 (s, 2H), 6.79 (dd, J = 7.7 Hz, J = 0.7 Hz, 1H), 6.95 (d, J = 1.4 Hz, 2H), 7.11 (td, J = 7.7 Hz, J = 2.1 Hz 1H), 7.16−7.22 (m, 4H), 7.23 (t, $J = 1.4$ Hz, 1H), 7.31 (t, $J = 7.7$ Hz, 1H), 7.37 (t, $J = 7.7$ Hz, 2H); ¹³C NMR (100.6 MHz, CDCl₃) δ 28.0, 31.5, 34.9, 40.0, 46.4, 119.4, 120.4, 125.8, 126.3, 126.4, 127.9, 129.9, 130.1, 131.1, 136.5, 142.7, 143.3, 148.3, 149.7; MS (EI) m/z (%) 398 (M⁺, 1.6), 231 (100); HRMS (EI) found 398.2974, calcd for $C_{30}H_{38}$ 398.2974.

1-Methoxy-2-[2-methyl-2-(3,5-di-tert-butylphenyl)propyl] benzene (4ee) (Table 5, Entry 7). Following the general procedure, a mixture of 2,4,6-tri-tert-butylbromobenzene (1e) (324.5 mg, 1.00 mmol), 2-methoxyphen[ylb](#page-2-0)oronic acid (2e) (228.7 mg, 1.51 mmol), $K_3PO_4·H_2O$ (705.0 mg, 3.06 mmol), and a dioxane solution (480 μ L, 1.11 mM, 0.53 μ mol) of the precatalyst composed of Pd(dba)₂, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:5.0:23.3 ratio in dioxane (2 mL) was stirred at 100 °C for 13 h. The resulting reaction mixture was diluted with n-hexane (5 mL), filtered through a thin pad of silica gel (AcOEt), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel (0:1 to 1:10 to 1:0 AcOEt/n-hexane) to provide the title compound (283.0 mg, 0.803 mmol, 80%) as a colorless oil. ¹H NMR (699.8 MHz, CDCl₃) δ 1.30 (s, 18H), 1.33 (s, 6H), 2.88 (s, 2H), 3.63 (s, 3H), 6.65 $(dd, J = 7.0 \text{ Hz}, J = 1.4 \text{ Hz}, 1H), 6.72 \text{ (td, } J = 7.0 \text{ Hz}, J = 0.7 \text{ Hz} 1H),$ 6.78 (d, J = 8.4 Hz, 1H), 7.1−7.15 (m, 3H), 7.24 (t, J = 1.4 Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ 28.0, 31.6, 34.9, 39.5, 43.7, 54.9, 110.0, 119.3, 119.5, 120.4, 127.0, 127.7, 132.0, 148.4, 149.5, 158.0; MS (EI) m/z (%) 352 (M⁺, 3.7), 231 (100); HRMS (EI) found 352.2765, calcd for $C_{25}H_{36}O$ 352.2766.

1-Ethoxy-2-[2-methyl-2-(3,5-di-tert-butylphenyl)propyl] benzene (4ef) (Table 5, Entry 8). Following the general procedure, a mixture of 2,4,6-tri-tert-butylbromobenzene (1e) (324.6 mg, 1.00 mmol), 2-ethoxyphenylboronic acid (2f) (249.3 mg, 1.50 mmol), $K_3PO_4·H_2O$ (715.7 mg, [3](#page-2-0).11 mmol), and a dioxane solution (495 μ L, 1.05 mM, 0.52 μ mol) of the precatalyst composed of Pd(dba)₂, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:5.0:24.5 ratio in dioxane (2 mL) was stirred at 100 °C for 9 h. The resulting reaction mixture was diluted with n-hexane (5 mL), filtered through a thin pad of silica gel (AcOEt), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel (10:1 AcOEt/n-hexane) to provide the title compound (352.4 mg, 0.961 mmol, 96%) as a colorless oil. ¹H NMR (399.9 MHz, CDCl₃) δ 1.31 (s, 18H), 1.38 (t, J = 6.8 Hz, 3H), 2.91 (s, 2H), 3.94 (q, J = 6.8 Hz, 2H), 6.66 (dd, J = 7.6 Hz, J = 2.0 Hz, 1H), 6.72 (td, J = 7.2 Hz, J = 1.2 Hz, 1H), 6.79 (d, $J = 8.0$ Hz, 1H), 7.12 (ddd, $J = 8.4$ Hz, $J = 7.6$ Hz, J = 2.0 Hz, 1H), 7.19 (d, J = 1.6 Hz, 2H), 7.26 (t, J = 1.6 Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ 15.0, 27.8, 31.6, 34.9, 39.6, 43.5, 63.1, 110.8, 119.3, 119.4, 120.3, 127.0, 127.7, 132.1, 149.0, 149.6, 157.5; MS (EI) m/z (%) 366 (M⁺, 4.4), 231 (100); HRMS (EI) found 366.2923, calcd for C₂₆H₃₈O 366.2923.

1-Isopropoxy-2-[2-methyl-2-(3,5-di-tert-butylphenyl) propyl]benzene (4eg) (Table 5, Entry 9). Following the general procedure, a mixture of 2,4,6-tri-tert-butylbromobenzene (1e) (324.9 mg, 1.00 mmol), 2-isopropoxyphenylboronic acid (2g) (269.7 mg, 1.50 mmol), $K_3PO_4·H_2O$ (733.[6](#page-2-0) mg, 3.19 mmol), and a dioxane solution (480 μ L, 1.05 mM, 0.50 μ mol) of the precatalyst composed of Pd(dba)₂, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:4.7:23.3 ratio in dioxane (2 mL) was stirred at 100 °C for 13 h. The resulting reaction mixture was diluted with n -hexane (5 mL), filtered through a thin pad of silica gel (AcOEt), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel (1:30 to 1:40 AcOEt/n-hexane) to provide the title compound (374.9 mg, 0.985 mmol, 99%) as a colorless oil. ¹H NMR (399.9 MHz, CDCl₃) δ 1.3 (m, 30H), 2.89 (s, 2H), 4.51 (sept, J = 6.0 Hz, 1H), 6.65−6.75 (m, 2H), 6.81 (d, J = 8.4, 1H), 7.11 (ddd, $J = 7.6$ Hz, $J = 6.8$ Hz, $J = 2.0$ Hz, 1H), 7.22 (d, $J = 1.6$ Hz, 2H), 7.26 (m, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ 22.2, 27.8, 31.6, 34.9, 39.7, 43.6, 69.1, 112.0, 119.1, 119.3, 120.3, 126.8, 128.5, 132.4, 149.1, 149.6, 156.2; MS (EI) m/z (%) 380 (M⁺, 3.9), 231 (100); HRMS (EI) found 380.3078, calcd for $C_{27}H_{40}O$ 380.3079.

4-Methyl-4′-methoxybiphenyl (3fi) ¹⁵ (Table 6, Entry 1). Following the general procedure, a mixture of 4-chloroanisole (1f) (143.0 mg, 1.00 mmol), 4-methylphenylb[oro](#page-11-0)nic acid (2i) (204.0 mg, 1.50 mmol), $K_3PO_4·H_2O$ (694.7 mg, 3.02 mmol), [an](#page-3-0)d a dioxane solution (200 μ L, 1.23 mM, 0.25 μ mol) of the precatalyst composed of $Pd(dba)₂$, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:4.1:21.6 ratio in dioxane (2 mL) was stirred at 100 °C for 180 min. The resulting reaction mixture was diluted with *n*-hexane (5 mL) , filtered through a thin pad of silica gel (1:5 AcOEt/n-hexane), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel $(1:5 \text{ AcOE}t/n$ hexane) to provide the title compound (195.4 mg, 0.986 mmol, 99%) as a white solid: mp = 106−107 $^{\circ}$ C; ¹H NMR (270.1 MHz, CDCl₃) δ

2.38 (s, 3H), 3.84 (s, 3H), 6.96 (d, J = 8.8 Hz, 2H), 7.22 (d, J = 7.8 Hz, 2H), 7.45 (d, J = 8.0 Hz, 2H), 7.51 (d, J = 9.0 Hz, 2H); ¹³C NMR (100.6 MHz, CDCl3) δ 20.9, 55.0, 114.0, 126.4, 127.8, 129.3, 133.5, 136.1, 137.8, 158.8; MS (EI) m/z (%) 198 (M⁺, 100), 183 (55), 155 (34); HRMS (EI) found 198.1043, calcd for $C_{14}H_{14}O$ 198.1045.

4-Methyl-2'-methoxybiphenyl (3gi)^{3a} (Table 6, Entry 2). Following the general procedure, a mixture of 2-chloroanisole (1g) (160.5 mg, 1.13 mmol), 4-methylphenylbo[ro](#page-11-0)nic acid (2i) (204.5 mg, 1.49 mmol), $K_3PO_4·H_2O$ (697.0 mg, 3.03 mmol), [and](#page-3-0) a dioxane solution (230 μ L, 1.23 mM, 0.28 μ mol) of the precatalyst composed of $Pd(dba)$, CyR-Phos, and ethyl 4-bromobenzoate in a 1.0:4.1:21.6 ratio in dioxane (2 mL) was stirred at 100 °C for 120 min. The resulting reaction mixture was diluted with n -hexane (5 mL), filtered through a thin pad of silica gel (1:5 AcOEt/n-hexane), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel $(1:5 \text{ AcOEt}/n-)$ hexane) to provide the title compound (189.1 mg, 0.954 mmol, 84%) as a white solid: mp = 79–81 °C; ¹H NMR (270.1 MHz, CDCl₃) δ 2.39 (s, 3H), 3.80 (s, 3H), 6.98 (d, $J = 8.3$ Hz, 1H), 7.03 (d, $J = 7.3$) Hz, 1H), 7.22 (d, J = 7.8 Hz, 2H), 7.28–7.36 (m, 2H), 7.43 (d, J = 8.0 Hz, 2H); ¹³C NMR (100.6 MHz, CDCl₃) δ 21.1, 55.3, 111.0, 120.7, 128.3, 128.6, 129.3, 130.5, 130.7, 135.5, 136.4, 156.4; MS (EI) m/z (%) 198 (M⁺ , 100), 183 (48), 168 (39); HRMS (EI) found 198.1045, calcd for $C_{14}H_{14}O$ 198.1045.

4-Methyl-2′,4′-dimethoxybiphenyl (3hi) (Table 6, Entry 4). Following the general procedure, a mixture of 1-chloro-2,4 dimethoxybenzene (1h) (169.4 mg, 0.98 mmol), 4-methylphenylboronic acid (2i) (204.3 mg, 1.50 mmol), K3PO4·H2O (7[01.](#page-3-0)0 mg, 3.04 mmol), and a dioxane solution (2.0 mL, 1.25 mM, 2.5 μ mol) of the precatalyst composed of $Pd(dba)_2$, CyR-Phos, and ethyl 4bromobenzoate in a 1.0:4.0:21.6 ratio in dioxane (2 mL) was stirred at 100 °C for 240 min. The resulting reaction mixture was diluted with n-hexane (5 mL), filtered through a thin pad of silica gel (1:5 AcOEt/ n-hexane), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel $(1:20 \text{ AcOE}t/n$ -hexane) to provide the title compound (192.6 mg) 0.844 mmol, 86%) as a colorless oil. ¹H NMR (399.9 MHz, CDCl₃) δ 2.38 (s, 3H), 3.79 (s, 3H), 3.85 (s, 3H), 6.53−6.58 (m, 2H), 7.17 (m, 3H), 7.39 (d, J = 8.4 Hz, 2H); ¹³C NMR (100.6 MHz, CDCl₃) δ 21.0, 55.2, 55.3, 98.8, 104.4, 123.4, 128.6, 129.2, 131.0, 135.3, 135.9, 157.3, 160.0; MS (EI) m/z (%) 228 (M⁺, 100), 213 (23), 198 (19); HRMS (EI) found 228.1150, calcd for $C_{15}H_{16}O_2$ 228.1150.

4-Amino-4'-methoxybiphenyl $(3i\bar{h})^{16}$ (Table 6, Entry 5). Following the general procedure, a mixture of 4-chloroaniline (1i) (133.4 mg, 1.05 mmol), 4-methoxyphen[ylbo](#page-11-0)ronic acid (2h) (231.0 mg, 1.52 mmol), $K_3PO_4·H_2O$ (700.4 mg, 3.04 mmol), [an](#page-3-0)d a dioxane solution (0.80 mL, 1.25 mM, 10 μ mol) of the precatalyst composed of $Pd(dba)₂$, CyR-Phos, and 1-bromo-4-tert-butylbenzene in a 1.0:4.0:25.6 ratio in dioxane (2 mL) was stirred at 100 °C for 370 min. The resulting reaction mixture was diluted with n -hexane (5 mL) , filtered through a thin pad of alumina gel (acetone), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel $(2.1 \text{ CHCl}_3/n$ -hexane) to provide the title compound (147.3 mg, 0.739 mmol, 70%) as a white solid: mp = 140−142 °C; ¹H NMR (399.9 MHz, CDCl₃) δ 3.68 (brs, 2H), 3.83 $(s, 3H)$, 6.74 (d, J = 8.8 Hz, 2H), 6.94 (d, J = 8.8 Hz, 2 H), 7.36 (d, J = 8.8 Hz, 2 H), $J = 7.45$ (d, $J = 8.8$ Hz, 2 H); ¹³C NMR (100.6 MHz, CDCl₃) δ 55.3, 114.0, 115.4, 127.3, 127.5, 131.2, 133.8, 145.3, 158.3; MS (EI) m/z (%) 199 (M+ , 100), 184 (89), 156 (40); HRMS (EI) found 199.0999, calcd for $C_{13}H_{13}N_1O$ 199.0997.

6-Naphthalen-1-yl-pyridin-2-ylamine (3jj) (Table 7, Entry 1). Following the general procedure, a mixture of 2-amino-6-chloropyridine (1j) (128.7 mg, 1.00 mmol), 1-naphtylboronic acid (2j) (259.4 mg, 1.51 mmol), $K_3PO_4·H_2O$ (708.3 mg, 3.08 mmol), a[nd](#page-3-0) a dioxane solution (0.80 mL, 1.20 mM, 0.96 μ mol) of the precatalyst composed of Pd(dba)₂, CyR-Phos, and bromobenzene in a 1.0:4.1:22.9 ratio in tert-amyl alcohol (1.4 mL) was stirred at 100 °C for 70 min. The resulting reaction mixture was diluted with n -hexane (5 mL), filtered through a thin pad of silica gel (AcOEt), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel $(1:3 \text{ AcOEt}/\text{CHCl}_3)$ to provide the title compound (220.1 mg, 1.00 mmol, 100%) as a white solid: mp = 99− 101 °C; ¹H NMR (270.1 MHz, CDCl₃) δ 4.56 (brs, 2H), 6.54 (d, J = 8.3 Hz, 1H), 6.91 (d, J = 7.3 Hz, 1H), 7.35−7.65 (m, 5H), 7.80−7.95 (m, 2H), 8.11 (d, J = 8.8 Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ 106.7, 114.6, 125.0, 125.5, 125.8, 125.9, 126.6, 127.9, 128.2, 130.9, 133.6, 137.7, 138.5, 157.1, 158.1; MS (EI) m/z (%) 220 (M⁺, 56), 219

(100); HRMS (EI) found 220.1002, calcd for $C_{15}H_{12}N_2$ 220.1000.
5-(4-Methoxyphenyl)-pyridin-2-ylamine (3kh)¹⁷ (Table 7, Entry 2). Following the general procedure, a mixture of 2-amino-5 chloropyridine (1k) (129.9 mg, 1.01 mmol), 4-methox[yph](#page-11-0)enylboronic acid (2h) (228.4 mg, 1.50 mmol), $K_3PO_4·H_2O$ (697.6 mg, 3.[03](#page-3-0) mmol), and a dioxane solution (0.80 mL, 1.20 mM, 0.96 μ mol) of the precatalyst composed of $Pd(dba)_{2}$, CyR-Phos, and bromobenzene in a 1.0:4.1:22.9 ratio in tert-amyl alcohol (1.4 mL) was stirred at 100 °C for 270 min. The resulting reaction mixture was diluted with n -hexane (5 mL) , filtered through a thin pad of silica gel $(0.1 \text{ to } 1.0 \text{ CHCl}_3/$ AcOEt), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel (3:2 $AcOEt/CHCl₃$) to provide the title compound (166.1 mg, 0.830) mmol, 82%) as a white solid: mp = 173–174 °C; ¹H NMR (270.1 MHz, CDCl₃) δ 3.84 (s, 3H), 4.47 (brs, 2H), 6.56 (d, J = 8.3 Hz, 1H), 6.96 (d, J = 8.8 Hz, 2H), 7.43 (d, J = 8.8 Hz, 2H), 7.62 (dd, J = 8.5 Hz, $J = 2.5$ Hz, 1H), 8.28 (d, $J = 1.7$ Hz, 1H); ¹³C NMR (100.6 MHz, CDCl3) δ 55.3, 108.5, 114.3, 127.1, 127.3, 130.9, 136.2, 146.0, 157.2, 158.8; MS (EI) m/z (%) 200 (M⁺, 100); 185 (92); HRMS (EI) found 200.0950, calcd for $C_{12}H_{12}N_2O$ 200.0950.

6-(4-Methoxyphenyl)-pyridin-3-ylamine (3lh) (Table 7, Entry 3). Following the general procedure, a mixture of 5-amino-2 chloropyridine (1l) (128.5 mg, 1.00 mmol), 4-methoxyphenylboronic acid (2h) (228.3 mg, 1.50 mmol), K3PO4·H2O (691.4 [mg](#page-3-0), 3.00 mmol), and a dioxane solution (0.80 mL, 1.25 mM, 1.0 μ mol) of the precatalyst composed of $Pd(dba)₂$, CyR-Phos, and bromobenzene in a 1.0:4.1:20.8 ratio in dioxane (1.4 mL) was stirred at 100 °C for 5 h. The resulting reaction mixture was diluted with *n*-hexane (5 mL) , filtered through a thin pad of alumina gel (acetone), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel $(3:1$ AcOEt/n-hexane) to provide the title compound (185.5 mg, 0.926 mmol, 93%) as a white solid: mp = 109−110 °C; ¹H NMR (399.9 MHz, CDCl₃) δ 3.70 (brs, 2H), 3.84 $(s, 3H)$, 6.96 (dd, J = 9.2 Hz, 2 H), 7.03 (dd, J = 8.4 Hz, J = 2.8 Hz, 1H), 7.47 (dd, $J = 8.4$ Hz, $J = 0.8$ Hz, 1H), 7.83 (d, $J = 8.8$ Hz, 2H), 8.15 (dd, J = 2.8 Hz, J = 0.4 Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ 55.3, 105.6, 107.9, 113.8, 128.0, 132.4, 150.0, 153.3, 157.9, 160.1; MS (EI) m/z (%) 200 (M+ , 100), 185 (75), 157 (63); HRMS (EI) found 200.0950, calcd for C₁₂H₁₂N₂O 200.0950.

6-(4-Methoxyphenyl)-pyridin-4-ylamine (3mh) (Table 7, Entry 4). Following the general procedure, a mixture of 4-amino-2 chloropyridine (1m) (129.7 mg, 1.01 mmol), 4-methoxyphenylbor-onic acid (2h) (227.2 mg, 1.50 mmol), K₃PO₄·H₂O (714.1 mg, 3.[10](#page-3-0)) mmol), and a dioxane solution (0.80 mL, 1.25 mM, 1.0 μ mol) of the precatalyst composed of $Pd(dba)_{2}$, CyR-Phos, and 1-bromo-4-tertbutylbenzene in a 1.0:4.0:22.9 ratio in dioxane (1.4 mL) was stirred at 100 °C for 6 h. The resulting reaction mixture was diluted with *n*hexane (5 mL), filtered through a thin pad of alumina gel (acetone), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel (acetone) to provide the title compound (97.6 mg, 0.487 mmol, 48%) as a white solid: mp = 137–139 °C; ¹H NMR (399.9 MHz, CDCl₃) δ 3.84 (s, 3H), 4.18 (brs, 2H), 6.44 (dd, J = 5.6 Hz, J = 2.0 Hz, 1H), 6.88 (d, J = 2.0 Hz, 1H), 6.96 (d, $J = 8.8$ Hz, 2H), 7.87 (d, $J = 8.8$ Hz, 2H), 8.27 (d, J = 5.6 Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ 55.1, 113.8, 119.9, 122.3, 127.0, 132.1, 136.7, 140.9, 147.5, 159.2; MS (EI) m/z (%) 200 (M+ , 100), 185 (46), 157 (28); HRMS (EI) found 200.0949, calcd for $C_{12}H_{12}N_2O$ 200.0950.

6-(2-Isopropoxyphenyl)-pyridin-5-ylamine (3 ng) (Table 7, Entry 5). Following the general procedure, a mixture of 3-amino-2 chloropyridine (1n) (129.4 mg, 1.01 mmol), 2-isopropoxyphenylboronic acid $(2g)$ $(279.1 \text{ mg}, K_3PO_4·H_2O$ $(721.4 \text{ mg}, 3.13 \text{ mmol})$, an[d a](#page-3-0) tert-amyl alcohol solution (0.80 mL, 1.20 mM, 0.96 μ mol) of the

precatalyst composed of $Pd(dba)_{2}$, CyR-Phos, and bromobenzene in a 1.0:4.1:23.8 ratio in tert-amyl alcohol (1.4 mL) was stirred at 100 °C for 70 min. The resulting reaction mixture was diluted with n -hexane (5 mL), filtered through a thin pad of silica gel (AcOEt), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel (1:2 AcOEt/ $CHCl₃$) to provide the title compound (225.6 mg, 0.988 mmol, 98%) as a white solid: mp = 87–89 °C; ¹H NMR (270.1 MHz, CDCl₃) δ 1.19 (d, $J = 6.1$ Hz, 6 H), 3.93 (brs, 2H), 4.38 (sept, $J = 6.1$ Hz, 1H), 6.95−7.20 (m, 4H), 7.35 (ddd, J = 8.0 Hz, J = 7.3 Hz, J = 1.7 Hz, 1H), 7.44 (dd, $J = 7.6$ Hz, $J = 1.5$ Hz, 1H), 8.14 (dd, $J = 4.4$ Hz, $J = 1.7$ Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ 21.6, 71.8, 115.7, 121.7, 122.1, 122.4, 129.1, 129.8, 131.7, 139.0, 141.0, 144.0, 154.4; MS (EI) m/z (%) 228 (M⁺ , 21), 213 (36), 185 (100), 169 (57); HRMS (EI) found 228.1263, calcd for C₁₄H₁₆N₂O 228.1263.

2,6-Dimethyl-5-(2-methylphenyl)-pyrimidin-4-ylamine (3oa)12d (Table 7, Entry 6). Following the general procedure, a mixture of 5-chloro-2,6-dimethyl-4-pyrimidinamine (1o) (158.6 mg, 1.01 [mmo](#page-11-0)l), 2-methylphenylboronic acid (2a) (205.8 mg, 1.51 mmol), $K_3PO_4·H_2O$ (700.[4 m](#page-3-0)g, 3.04 mmol), and a dioxane solution (0.80 mL, 1.25 mM, 1.0 μ mol) of the precatalyst composed of Pd(dba)₂, CyR-Phos, and 1-bromo-4-tert-butylbenzene in a 1.0:4.1:22.4 ratio in dioxane (1.4 mL) was stirred at 100 °C for 6 h. The resulting reaction mixture was diluted with n-hexane (5 mL), filtered through a thin pad of silica gel (acetone), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel (7:1 acetone/CHCl₃) to provide the title compound (155.1 mg, 0.727 mmol, 72%) as a white solid: mp = 204-206 °C; ¹H NMR (399.9 MHz, CDCl3) δ 2.07 (s, 3H), 2.11 (s, 3H), 2.53 (s, 3H), 4.65 (brs, 2H), 7.11 (d, J = 6.4 Hz, 1H), 7.28–7.35 (m, 3H); ¹³C NMR (100.6 MHz, CDCl3) δ 19.2, 21.7, 25.5, 113.5, 126.8, 128.5, 129.8, 130.7, 133.6, 137.1, 160.9, 162.5, 165.7; MS (EI) m/z (%) 213 (M⁺ , 80), 198 (100); HRMS (EI) found 213.1265, calcd for $C_{13}H_{15}N_3$ 213.1266.

2-Hydroxy-6-(4-methoxyphenyl)pyridine (3ph) (Table 7, Entry 7). Following the general procedure, a mixture of 6-chloro-2 hydroxypyridine (1p) (131.6 mg, 1.02 mmol), 4-methoxyphenylboronic acid $(2h)$ (229.9 mg, 1.51 mmol), $K_3PO_4·H_2O$ (705.8 mg, 3.[06](#page-3-0)) mmol), and a dioxane solution (0.80 mL, 1.20 mM, 0.96 μ mol) of the precatalyst composed of $Pd(dba)_2$, CyR-Phos, and 1-bromo-4-tertbutylbenzene in a 1.0:4.1:19.2 ratio in dioxane (1.4 mL) was stirred at 100 °C for 6.5 h. The resulting reaction mixture was diluted with nhexane (5 mL), filtered through a thin pad of silica gel (acetone), and concentrated under reduced pressure. The crude material obtained was purified by flash chromatography on silica gel (7:1 acetone/ $CHCl₃$) to provide the title compound (145.0 mg, 0.721 mmol, 71%) as a white solid: mp = 199–201 $^{\circ}$ C; ¹H NMR (399.9 MHz, CDCl₃) δ 3.86 (s, 3H), 6.42 (dd, $J = 7.2$ Hz, $J = 1.2$ Hz, 1H), 6.49 (dd, $J = 8.8$ Hz, J = 0.8 Hz, 1H), 7.01 (d, J = 9.2 Hz, 2H), 7.46 (dd, J = 8.8 Hz, J = 7.2 Hz, 1H), 7.66 (d, J = 8.8 Hz, 2H), 11.2−12.7 (brs, 1H); 13C NMR (100.6 MHz, CDCl3) δ 55.4, 103.9, 114.6, 117.7, 125.9, 128.0, 141.4, 146.7, 161.1, 165.1; MS (EI) m/z (%) 201 (M⁺ , 100), 173 (26), 158 (58), 130 (21); HRMS (EI) found 201.0790, calcd for $C_{12}H_{11}N_1O_2$ 201.0790.

■ ASSOCIATED CONTENT

3 Supporting Information

Full spectroscopic data including ${}^{1}\mathrm{H}$ NMR, ${}^{13}\mathrm{C}$ NMR, and ${}^{31}\mathrm{P}$ NMR spectra of all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

■ AUTHOR INFORMATION

Corresponding Author

*E-mail: hoshi@gs.niigata-u.ac.jp.

Notes

The auth[ors declare no competin](mailto:hoshi@gs.niigata-u.ac.jp)g financial interest.

■ ACKNOWLEDGMENTS

This work was supported financially by a Grant-in-Aid for Scientific Research (21550100) from MEXT.

■ REFERENCES

(1) (a) Nishiyama, M.; Yamamoto, T.; Koie, Y. Tetrahedron Lett. 1998, 39, 617. (b) Yamamoto, T.; Nishiyama, M.; Koie, Y. Tetrahedron Lett. 1998, 39, 2367.

(2) For recent reviews, see: (a) Littke, A. F.; Fu, G. C. Angew. Chem., Int. Ed. 2002, 41, 4176. (b) Miura, M. Angew. Chem., Int. Ed. 2004, 43, 2201. (c) Christmann, U.; Vilar, R. Angew. Chem., Int. Ed. 2005, 44, 366. (d) Kantchev, E. A. B.; O'Brien, C. J.; Organ, M. G. Angew. Chem., Int. Ed. 2007, 46, 2768. (e) Marion, N.; Nolan, S. P. Acc. Chem. Res. 2008, 41, 1440. (f) Martin, R.; Buchwald, S. L. Acc. Chem. Res. 2008, 41, 1461. (g) Würtz, S.; Glorius, F. Acc. Chem. Res. **2008**, 1, 1523. (h) Hartwig, J. F. Acc. Chem. Res. 2008, 41, 1534. (i) Fu, G. C. Acc. Chem. Res. 2008, 41, 1555. (j) Surry, D. S.; Buchwald, S. L. Angew. Chem., Int. Ed. 2008, 47, 6338. (k) Molander, G. A.; Canturk, B. Angew. Chem., Int. Ed. 2009, 48, 9240. (l) Fleckenstein, C. A.; Plenio, H. Chem. Soc. Rev. 2010, 39, 694. (m) So, C. M.; Kwong, F. Y. Chem. Soc. Rev. 2011, 40, 4963. (n) Fortman, G. C.; Nolan, S. P. Chem. Soc. Rev. 2011, 40, 5151. (o) Valente, C.; Çalimsiz, S.; Hoi, K. H.; Mallik, D.; Sayah, M.; Organ, M. G. Angew. Chem., Int. Ed. 2012, 51, 3314. (3) For some examples of effective phosphine ligands, see: (a) Old, D. W.; Wolfe, J. P.; Buchwald, S. L. J. Am. Chem. Soc. 1998, 120, 9722. (b) Kočovský, P.; Vyskočil, Š.; Císařová, I.; Sejbal, J.; Tišlerová, I.; Smrčina, M.; Lloyd-Jones, G. C.; Stephen, S. C.; Butts, C. P.; Murray, M.; Langer, V. J. Am. Chem. Soc. 1999, 121, 7714. (c) Shelby, Q.; Kataoka, N.; Mann, G.; Hartwig, J. J. Am. Chem. Soc. 2000, 122, 10718. (d) Kataoka, N.; Shelby, Q.; Stambuli, J. P.; Hartwig, J. F. J. Org. Chem. 2002, 67, 5553. (e) Huang, X.; Anderson, K. W.; Zim, D.; Jiang, L.; Klapars, A.; Buchwald, S. L. J. Am. Chem. Soc. 2003, 125, 6653. (f) Walker, S. D.; Barder, T. E.; Martinelli, J. R.; Buchwald, S. L. Angew. Chem., Int. Ed. 2004, 43, 1871. (g) Ackermann, L.; Born, R.; Spatz, J. H.; Meyer, D. Angew. Chem., Int. Ed. 2005, 44, 7216. (h) Iwasawa, T.; Komano, T.; Tajima, A.; Tokunaga, M.; Obora, Y.; Fujihara, T.; Tsuji, Y. Organometallics 2006, 25, 4665. (i) Ohta, H.; Tokunaga, M.; Obora, Y.; Iwai, T.; Iwasawa, T.; Fujihara, T.; Tsuji, Y. Org. Lett. 2007, 9, 89. (j) So, C. M.; Lau, C. P.; Kwong, F. Y. Org. Lett. 2007, 9, 2795. (k) Fleckenstein, C. A.; Plenio, H. Chem.-Eur. J. 2007, 13, 2701. (l) Fors, B. P.; Watson, D. A.; Biscoe, M. R.; Buchwald, S. L. J. Am. Chem. Soc. 2008, 130, 13552. (m) So, C. M.; Yeung, C. C.; Lau, C. P.; Kwong, F. Y. Org. J. Org. Chem. 2008, 73, 7803. (n) So, C. M.; Zhou, Z.; Lau, C. P.; Kwong, F. Y. Angew. Chem., Int. Ed. 2008, 47, 6402. (o) Iwasawa, T.; Kamei, T.; Watanabe, S.; Nishiuchi, M.; Kawamura, Y. Tetrahedron Lett. 2008, 49, 7430. (p) Watson, D. A.; Su, M.; Teverovsky, G.; Zhang, Y.; García-Fortanet, J.; Kinzel, T.; Buchwald, S. L. Science 2009, 325, 1661. (q) Hicks, J. D.; Hyde, A. M.; Cuezva, A. M.; Buchwald, S. L. J. Am. Chem. Soc. 2009, 131, 16720. (r) Fujihara, T.; Yoshida, S.; Terao, J.; Tsuji, Y. Org. Lett. 2009, 11, 2121. (s) Lundgren, R. J.; Peters, B. D.; Alsabeh, P. G.; Stradiotto, M. Angew, Chem. Int. Ed. 2010, 49, 4071. (t) Tang, W.; Capacci, A. G.; Wei, X.; Li, W.; White, A.; Patel, N. D.; Savoie, J.; Gao, J. J.; Rodriduez, S.; Qu, B.; Haddad, N.; Lu, B. Z.; Krishnamurthy, D.; Yee, N. K.; Senanayake, C. H. Angew. Chem., Int. Ed. 2010, 49, 5879. (u) Wu, X.; Fors, B. P.; Buchwald, S. L. Angew. Chem., Int. Ed. 2011, 50, 9943. (v) To, S. C.; Kwong, F. Y. Chem. Commun. 2011, 47, 5079. (w) Yang, Y.; Buchwald, S. L. J. Am. Chem. Soc. 2013, 135, 10642.

(4) For some examples of effective carbene ligands, see: (a) Navarro, O.; Kekky, R. A.; Nolan, S. P. J. Am. Chem. Soc. 2003, 125, 16194. (b) Altenhoff, G.; Goddard, R.; Lehmann, C. W.; Glorius, F. Angew. Chem., Int. Ed. 2003, 42, 3690. (c) Altenhoff, G.; Goddard, R.; Lehmann, C. W.; Glorius, F. J. Am. Chem. Soc. 2004, 126, 15195. (d) Fujihara, T.; Yoshida, S.; Ohta, H.; Tsuji, Y. Angew. Chem., Int. Ed. 2008, 47, 8310. (e) Çalimsiz, S.; Sayah, M.; Mallik, D.; Organ, M. G. Angew. Chem., Int. Ed. 2010, 49, 2014. (f) Peh, G.-R.; Kantchev, E. A. B.; Er, J.-C.; Ying, J. Y. Chem.—Eur. J. 2010, 16, 4010. (g) Tu, T.; Sun, Z.; Fang, W.; Xu, M.; Zhou, Y. Org. Lett. 2012, 14, 4250.

(5) (a) Hoshi, T.; Nakazawa, T.; Saitoh, I.; Mori, A.; Suzuki, T.; Sakai, J.; Hagiwara, H. Org. Lett. 2008, 10, 2063. (b) Hoshi, T.; Saitoh, I.; Nakazawa, T.; Suzuki, T.; Sakai, J.; Hagiwara, H. J. Org. Chem. 2009, 74, 4013.

(6) Caron, S.; Massett, S. S.; Bogle, D. E.; Castaldi, M. J.; Braish, T. F. Org. Process Res. Dev. 2001, 5, 254.

 (7) We added an aryl bromide with the expectation that the Pd⁰ precatalyst was converted in situ into the Pd^H precatalyst. However, no evidence for the generation of the Pd^{II} precatalyst was observed by ^{31}P NMR analysis. For some efficient Pd^{II} precatalysts, see: (a) Biscoe, M. R.; Fors, B. P.; Buchwald, S. L. J. Am. Chem. Soc. 2008, 130, 6686. (b) Kinzel, T.; Zhang, Y.; Buchwald, S. L. J. Am. Chem. Soc. 2010, 132, 14073. (c) Chartoire, A.; Lesieur, M.; Slawin, A. M. Z.; Nolan, S. P.; Cazin, C. S. J. Organometallics 2011, 30, 4432.

(8) Lee, D.-H.; Jin, M.-J. Org. Lett. 2011, 13, 252.

(9) To the best of our knowledge, this represents the first example of organic group exchange between organopalladium and organoborane. For other organometallic reagents that undergo organic group exchange with organopalladium, see the following. Organozinc: (a) van Asselt, R.; Elsevier, C. J. Organometallics 1994, 13, 1972. (b) Liu, Q.; Lan, Y.; Liu, J.; Li, G.; Wu, Y.-D.; Lei, A. J. Am. Chem. Soc. 2009, 131, 10201. (c) Çalimsiz, S.; Sayah, M.; Mallik, D.; Organ, M. G. Angew, Chem. Int. Ed. 2010, 49, 2014. Organosilicon: (d) Liang, Y.; Zhang, S.; Xi, Z. J. Am. Chem. Soc. 2011, 133, 9204. Organomagnesium: see ref 9a.

(10) (a) Barder, T. E.; Walker, S. D.; Martinelli, J. R.; Buchwald, S. L. J. Am. Chem. Soc. 2005, 127, 4685. (b) Pan, J.; Su, M.; Buchwald, S. L. Angew. Chem., Int. Ed. 2011, 50, 8647.

(11) Pozharskii, A. T.; Soldatenkov, A.; Katritzky, A. R. Heterocycles in Life and Society, 2nd ed.; John Wiley & Sons Ltd: West Sussex, 2003. (12) (a) Itoh, T.; Sato, K.; Mase, T. Adv. Synth. Catal. 2004, 346, 1859. (b) Anderson, K. W.; Buchwald, S. L. Angew. Chem., Int. Ed. 2005, 44, 7173. (c) Itoh, T.; Mase, T. Tetrahedron Lett. 2005, 46, 3573. (d) Billingsley, K. L.; Anderson, K. W.; Buchwald, S. L. Angew. Chem., Int. Ed. 2006, 45, 3484. (e) Guram, A. S.; King, A. O.; Allen, J. G.; Wang, X.; Schenkel, L. B.; Chan, J.; Bunel, E. E.; Faul, M. M.; Larsen, R. D.; Martinelli, M. J.; Reider, P. J. Org. Lett. 2006, 8, 1787. (f) Billingsley, K. L.; Buchwald, S. L. J. Am. Chem. Soc. 2007, 129, 3358. (g) Guram, A. S.; Wang, X.; Bunel, E. E.; Faul, M. M.; Larsen, R. D.; Martinelli, M. J. J. Org. Chem. 2007, 72, 5104. (h) Fleckenstein, C. A.; Plenio, H. Chem.-Eur. J. 2008, 14, 4267. (i) Whelligan, D. K.; Thomson, D. W.; Taylor, D.; Hoelder, S. J. Org. Chem. 2010, 75, 11. (13) Knorr, R.; Pires, C.; Behringer, C.; Menke, T.; Freudenreich, J.;

Rossmann, E. C.; Böhrer, P. J. Am. Chem. Soc. 2006, 128, 14845. (14) Wolfe, J. P.; Singer, R. A.; Yang, B. H.; Buchwald, S. L. J. Am. Chem. Soc. 1999, 121, 9550.

(15) Lebel, H.; Janes, M. K.; Charette, A. B.; Nolan, S. P. J. Am. Chem. Soc. 2004, 126, 5046.

(16) Razler, T. M.; Hsiao, Y.; Qian, F.; Fu, R.; Khan, R. K.; Doubleday, W. J. Org. Chem. 2009, 74, 1381.

(17) Heitman, L. H.; van Veldhoven, J. P. D.; Zweemer, A. M.; Ye, K.; Brussee, J.; IJzerman, A. P. J. Med. Chem. 2008, 51, 4724.